

Fast Pyrolysis of Lignocellulosic Biomass - a Versatile Tool for Making Chemicals



vTI
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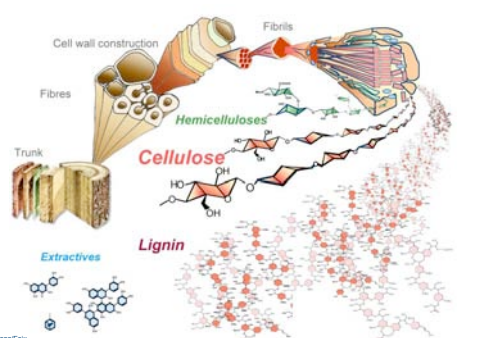
DBU Osnabrück, Germany
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Dietrich Meier

Content

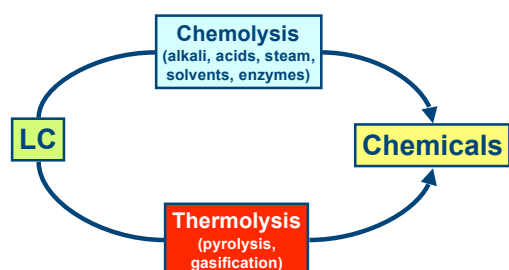
- Fundamentals
- Production processes / biorefinery concepts
- Separation options
- Upgrading options
- „zero waste“ concept

Lignocellulose Constituents (Wood)



after Hoffmann/Falk
modified by Lehtonen

From Lignocelluloses to Chemicals

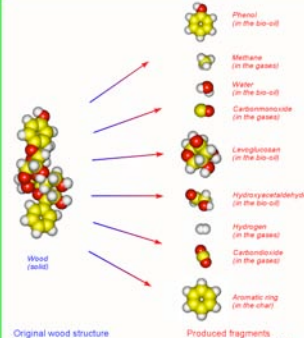


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    graph TD
      LC[LC] --> Chemolysis[Chemolysis  
(alkali, acids, steam,  
solvents, enzymes)]
      Chemolysis --> Chemicals[Chemicals]
      Chemicals --> Thermolysis[Thermolysis  
(pyrolysis,  
gasification)]
      Thermolysis --> LC
  
```

Fast Pyrolysis Principle

- Fast chemical degradation due to rapid heating in the absence of oxygen
- Process characteristics:
 - Temperature 500 °C
 - Pressure 1 bar
 - Particle size < 5 mm
 - t vapours < 2 s
- The main product is a liquid: Bio-Oil or Bio Crude Oil (BCO; approx. 70 wt.%)



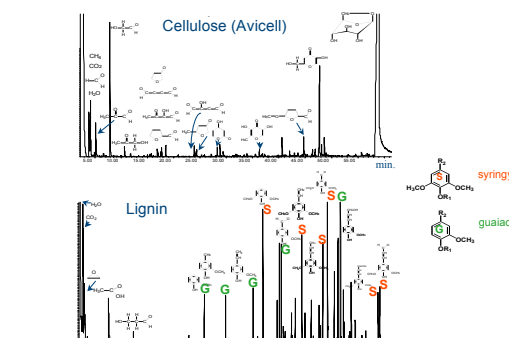
Original wood structure (at 20 °C)

Produced fragments (after pyrolysis at 500 °C)

- Phenol (in the bio-oil)
- Methane (in the gases)
- Water (in the bio-oil)
- Carbon monoxide (in the gases)
- Levoglucosan (in the bio-oil)
- Hydroxyacetalddehyde (in the bio-oil)
- Hydrogen (in the gases)
- Carbon dioxide (in the gases)
- Aromatic ring (in the char)

Prins, W., 2007

Products of Fast Pyrolysis from Py-GC/MS



Cellulose (Avicell)

Lignin

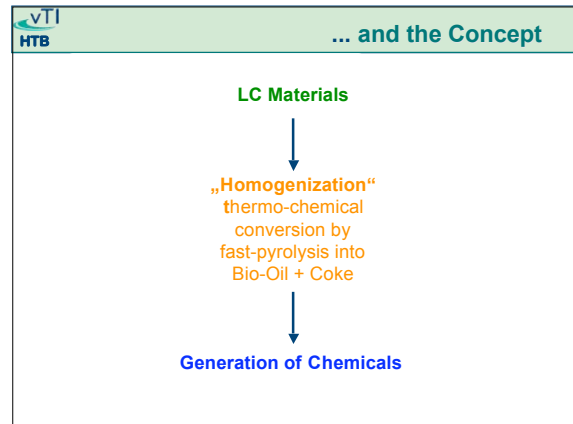
syringyl

guaiacyl

RT (min)

Biorefinery Concepts

- General
 - Use of lignocelluloses (LC): wood, residues (e.g. straw, bagasse)
 - no competition with food
 - restricted accessibility of biopolymers => pretreatment => conversion
- Classic approach
 - **limited feedstock selectivity** (conditioning)
 - (pretreatment necessary with **pressure processes**, e.g. organosolv, steam)
 - separation & cleaning
 - Use **only after conversion** or modification (degradation to monomers, functionalisation, polymerisation)
- Thermochemical approach
 - **broad selection of raw LC materials** (e.g. straw, bark, DDGs, shells, etc.)
 - simple thermal treatment by fast-pyrolysis at **atmospheric pressure**
 - **decentral conversion - central refining** (separation & cleaning)
 - **Direct or indirect use** after modification

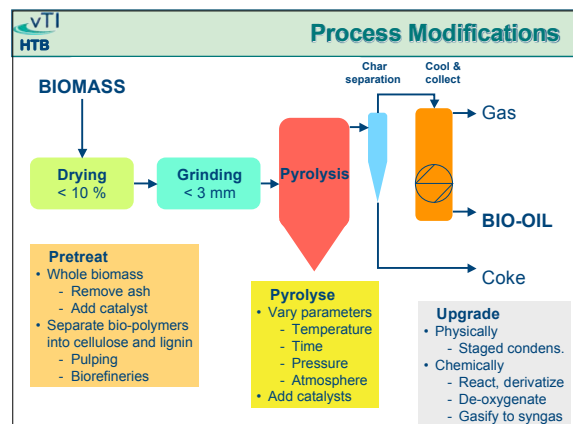
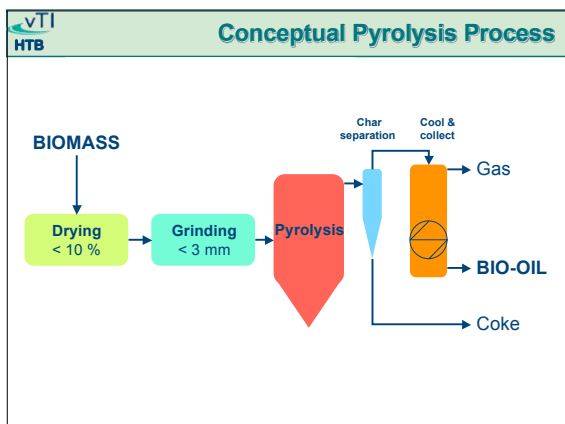
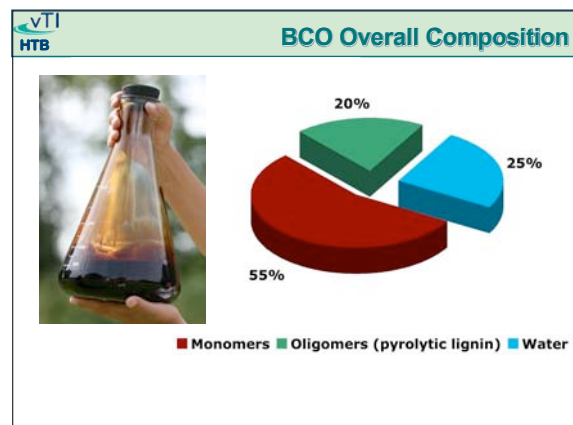


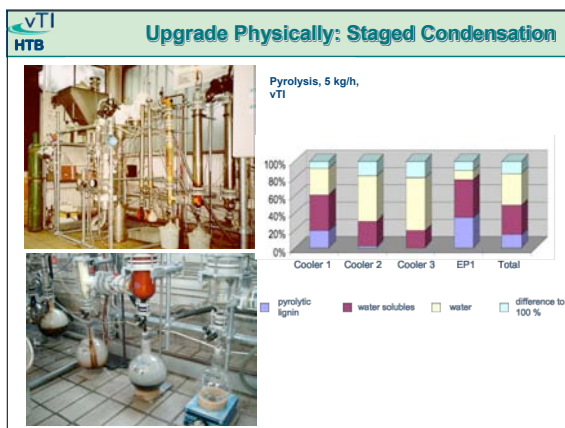
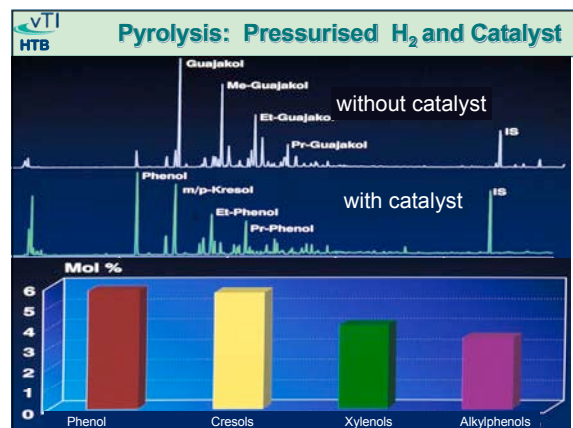
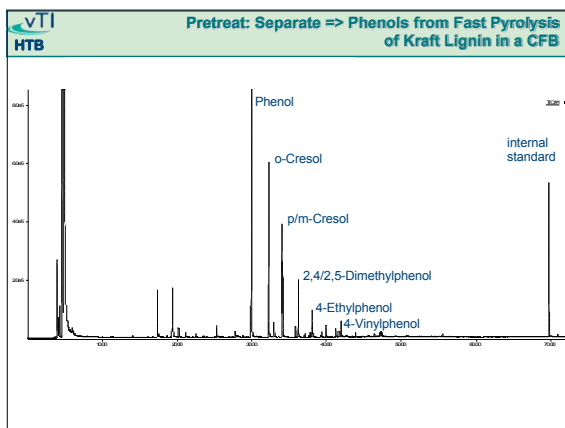
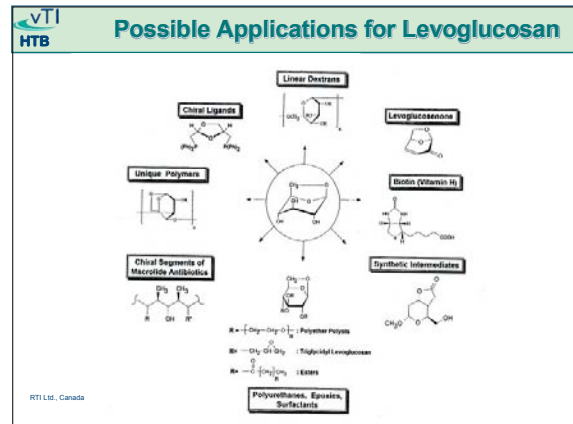
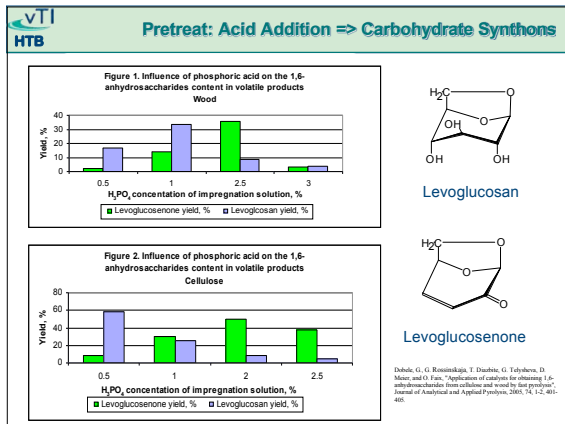
BCO Properties

Analysis	Pyrolysis liquids	Light fuel oil (Tempera 15)
Water, wt%	20-30	0.025
Solids, wt%	0.01-1	0
Ash, wt%	0.01-0.2	0.01
Nitrogen, wt%	0-0.4	0
Sulfur, wt%	0-0.05	0.2
Stability	Unstable	Stable
Viscosity (40 °C), cSt	15-35	3.0-7.5
Density (15 °C), kg/dm ³	1.10-1.30	0.89
Flash point, °C	40-110	60
Pour point, °C	-9- -36	-15
LHV, MJ/kg	13-18	40.3
pH	2-3	Neutral
Distillability	Not distillable	160-400 °C

1 MJ Feedstock 1 MJ Product

Prins, W., 2007





Upgrading Options for BCO

- **Reaction**
 - Add NH₃, urea => Slow N-release fertilizer
 - Add lime => BioLime™ => NOx/SO₂ reduction
- **Separation**
 - Add water => Flavour, Adhesives
 - Add high pressure steam (HPTT) => oxygenates, tar
- **Cracking**
 - Synthesis gas => FT-Diesel, MeOH, DME

