

‘NEUE CHEMISCHE PROZESSFENSTER - NOVEL PROCESS WINDOWS’

**Kick-Off-Meeting Forschungscluster
“Novel Process Windows”**

Osnabrück, 21.02.08

Volker Hessel

v.hessel@tue.nl; hessel@imm-mainz.de

**¹ Eindhoven University of Technology
Department of Chemical Engineering and Chemistry**

TU/e

**² Institut für Mikrotechnik Mainz GmbH
Chemical Micro and Milli Process Technologies**

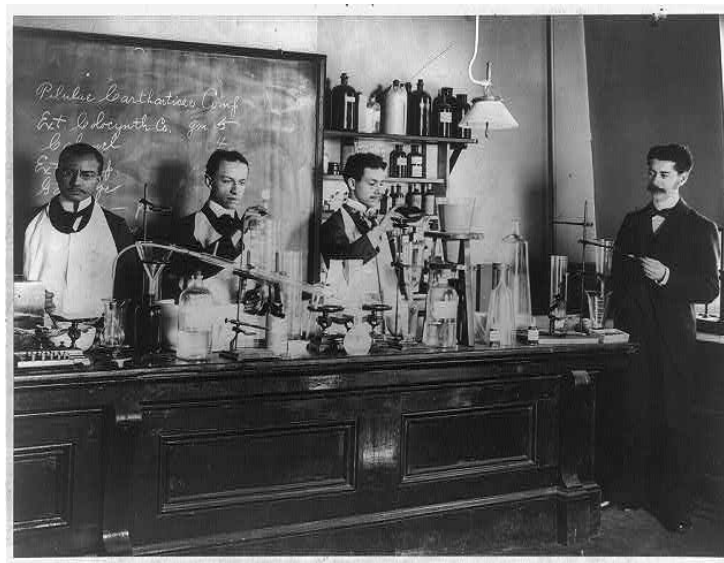
imm

- **Starting position**
- **From concept to research cluster**
 - **“Novel Process Windows”**
- **International reception**
 - **own related activities**

- **Starting Position**
 - **Novel Tools! Novel Chemistry?**



18-19th century



~1900

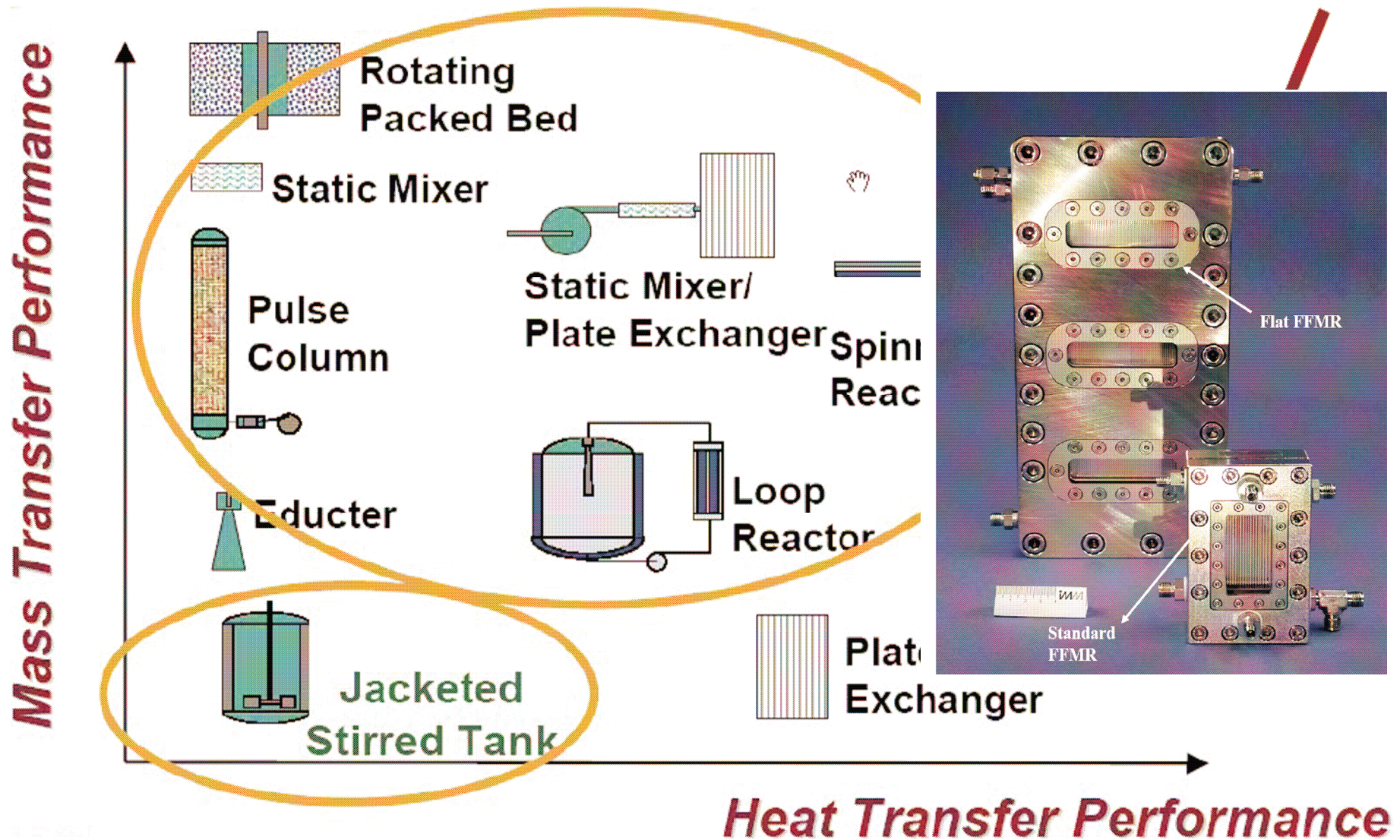


Today



Batch
procedure

The Continuous Quest for Still Better Reactors, Even Smarter Catalyst Packings, and More Intensified Processes



(Source: Rhodia)

Calculation of energy savings – Results per sector – Fine Chemical

||| A2

Results calculation energy savings estimates for Fine Chemical & Pharma sector

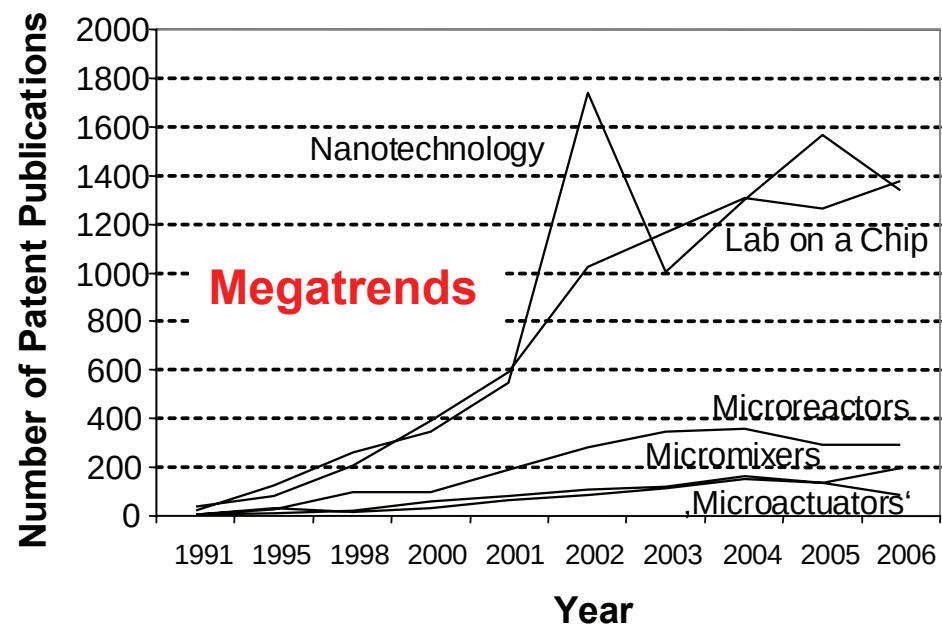
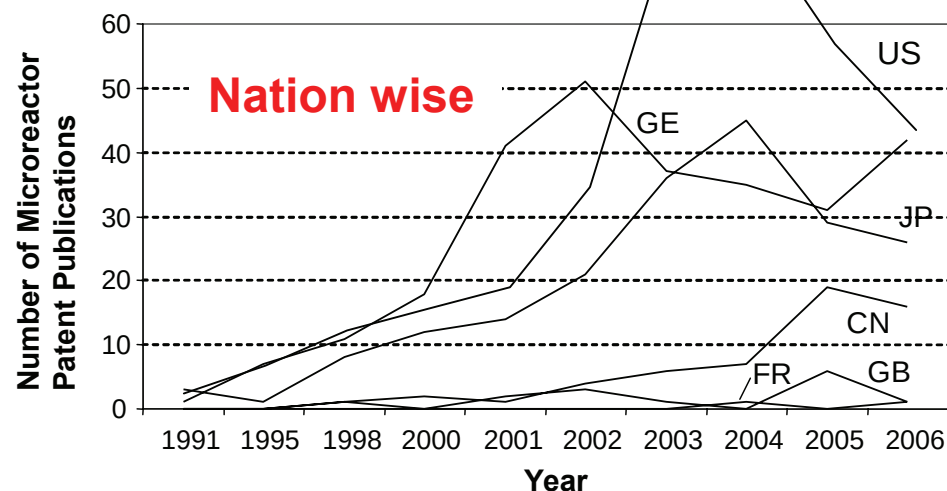
Fine Chem & Pharma		Multi-functional equipment	Micro/milli reactors	Microwave	High gravity fields	Total
Ultimate saving potential	PJ					
- Low		0	5	2	1	8
- High		1	7	3	3	14
penetration 2015		0%	8%	3%	20%	
penetration 2030		70%	88%	71%	100%	
Range 2015	low	0,0	0,4	0,1	0,2	0,7
	high	0,0	0,6	0,1	0,6	1,3
Range 2030	low	0,0	4,4	1,4	1,0	6,8
	high	0,7	6,2	2,1	3,0	12,0
Rate of introduction:						
[n] Introduction new tech.	year	10	12	15	5	
[m] plant/process replacement	year	15	15	15	15	
[y] Year of introduction		2015	2010	2012	2010	

**Building a Business Case on
Process Intensification, July 2006**
**Presented to Platform for Chain
Efficiency (PKE); Full Report**

**"The question
of whether
microreactors are
going to be used in
the future, I think
this is already
answered 'yes'." —
Dominique Roberge**

Nature 442, 7101 (2006) 351-352

- **Micro-/millireactors owe highest potential for global energy saving and competitiveness of chemical industry**
 - **Focus is on process intensification („milli“ and not „micro“, where sufficient)**
- >> Coming report (2008): European Process Intensification Roadmap, Stichting Energietransitie Ketenefficiency, The Netherlands**



Top-10 Patentees

Ran king	Patent Publications	Patentee
1	144	MERCK PATENT GMBH
2	135	BATTELLE MEMORIAL INSTITUTE
3	129	VELOCYS INC.
4	54	FORSCHUNGSZENTRUM KARLSRUHE GMBH
5	50	UHDE GMBH
6	46	SIEMENS AKTIENGESELLSCHAFT
7	45	CASIO COMPUTER CO., LTD.
8	40	DEGUSSA AG
9	39	BAYER AG
9	39	INSTITUT FUER MIKROTECHNIK MAINZ GMBH
10	37	CLARIANT INTERNATIONAL LTD.

V. Hessel, C. Knobloch, H. Löwe, „Review on Patents in Microreactor and Micro Process Engineering”, *Rec. Pat. Chem. Eng.* **1** (2008) 1-16.

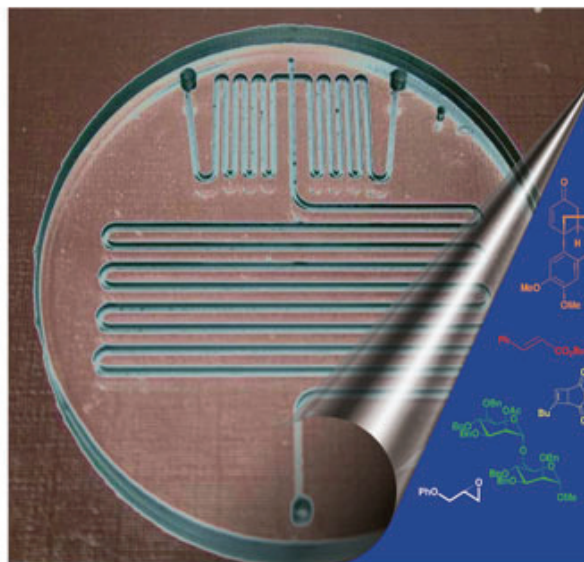
Hydrogen peroxide synthesis
Organolithium exchange
Chlorination
Dehydrogenation
Polyacrylate formation
Alkylation reactions based on
DAST Fluorination (Bayer)
Steroid Ozonolysis (Bayer)
Plasma reaction
Gas-liquid oxidation
PMMA manufacture
Grignard exchange reaction
Halogen-Lithium exchange
Swern-Moffat oxidation
Yellow nano pigments

Hessel, V., Löb, P., Löwe, H.
Microreactors for Organic Synthesis

Edited by Thomas Wirth

WILEY-VCH

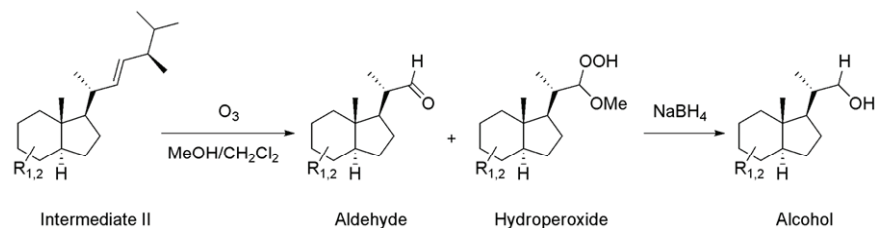
Microreactors in Organic Synthesis and Catalysis



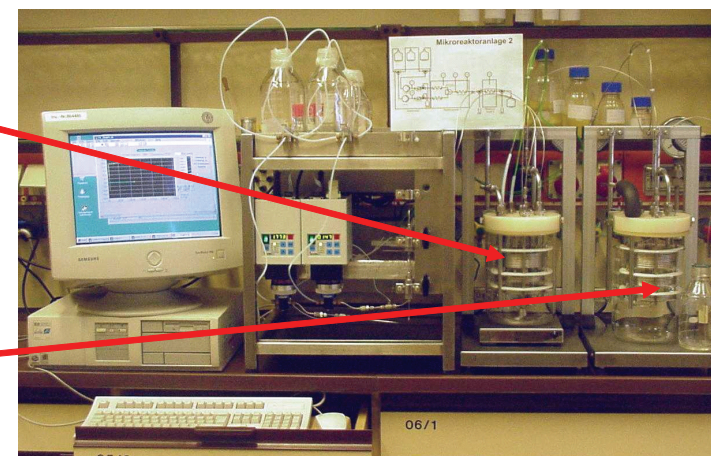
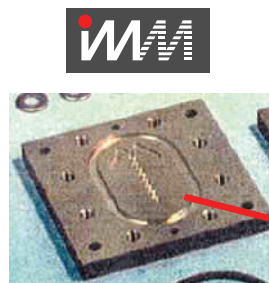
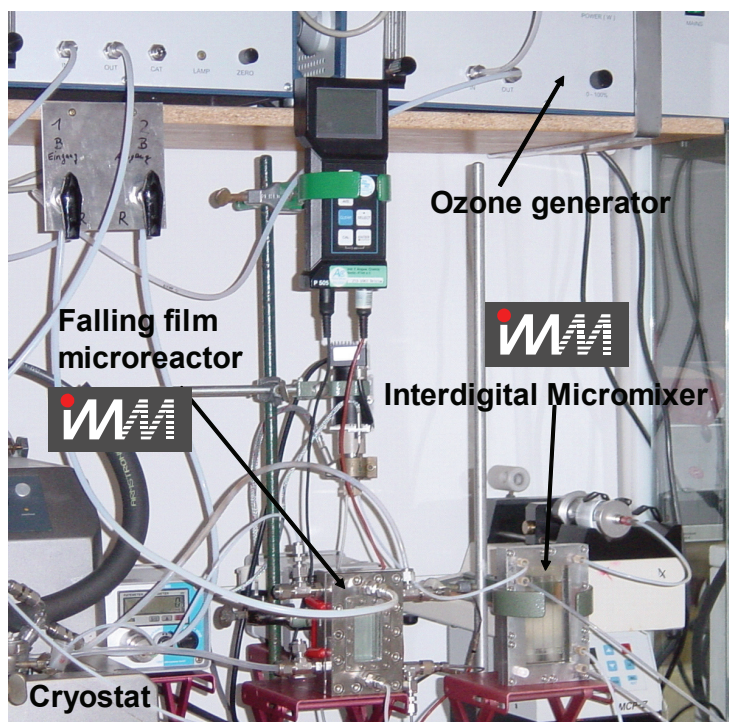
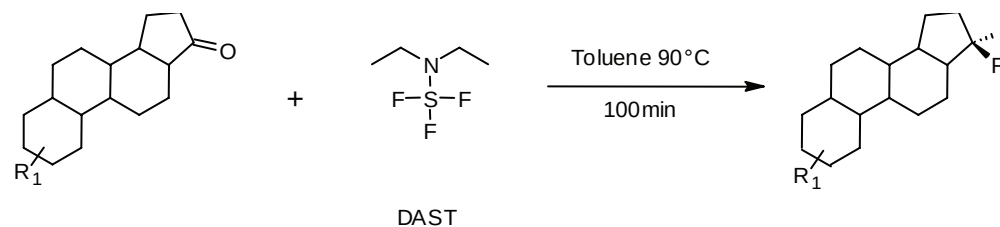
ion (Japanese company)
ylation (Japanese company)
oxidation to 2-methyl-1,4-
ne (Japanese company)
n of ethyl 3-oxobutanoate
(Sahi Glass)
ormation (Degussa / Uhde)
ate synthesis (Uhde)
natic compounds (Bristol-
yers-Squibb)
lymer intermediates (DSM
hemicals GmbH)
diaz pigments (Clariant)
roduction (Xi'an Company)
roduction (not disclosed)
enolate production (Merck)

up to Production", in: "Micro-
einheim (to be released 2008).

Ozonolysis



Fluorination

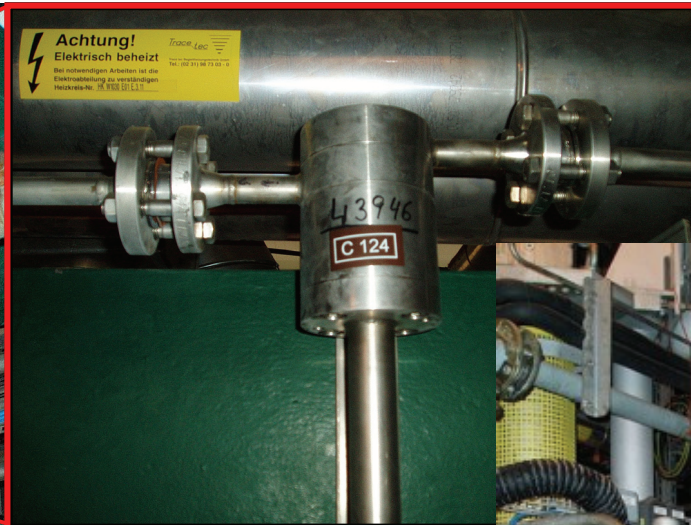


MICROSTRUCTURED REACTOR BASED PRODUCTION-PLANT



**micro
innova**

micro chemical engineering



D. Kirschneck, G. Tekautz,
Chem. Eng. Technol. 30, 3
(2007) 305-308.

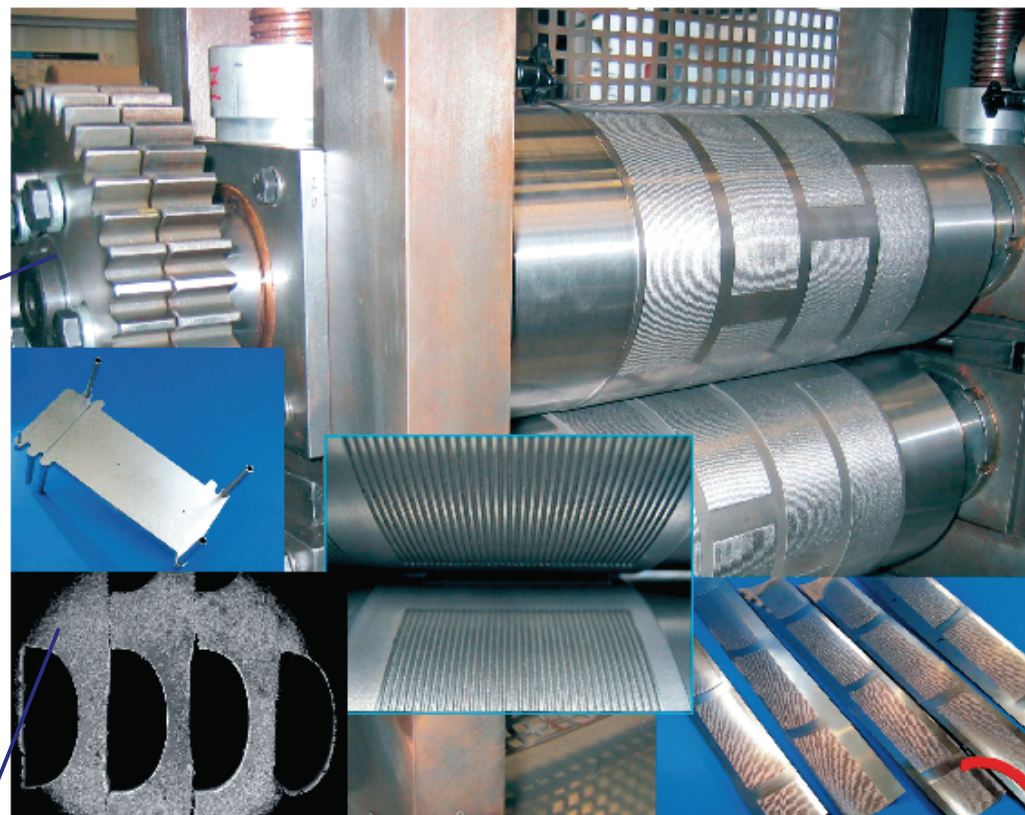
- In operation since June 2005
- Throughput 3 metric tons per hour; worldwide largest reported productivity by micro process engineering



MASS MANUFACTURE:

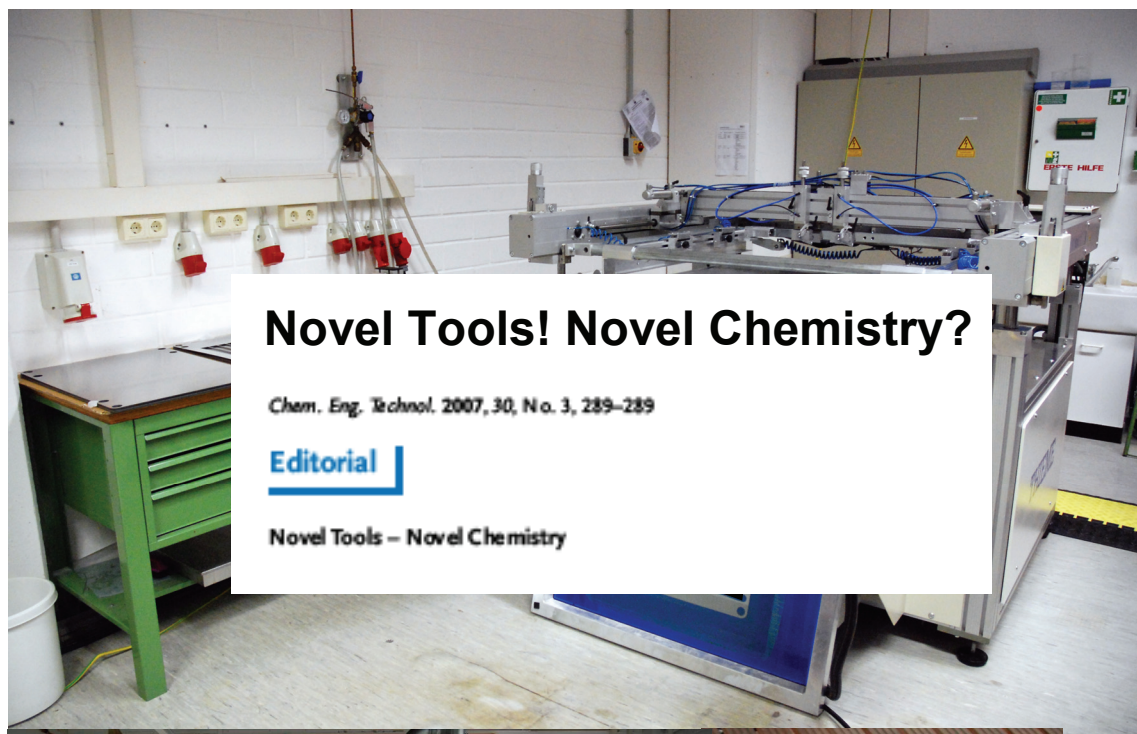
„Large Reactors“ for
Bulk Chemistry + Large lot sizes
for automotive + consumer good
industries

Rolling



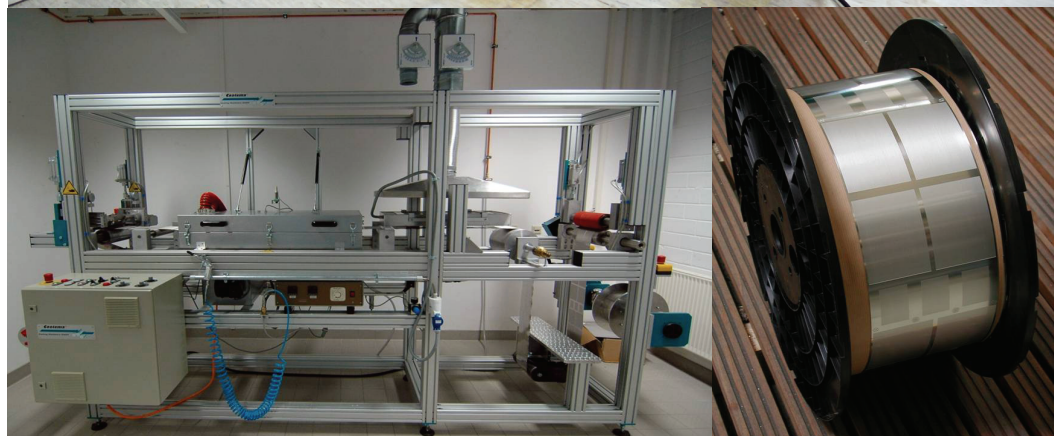
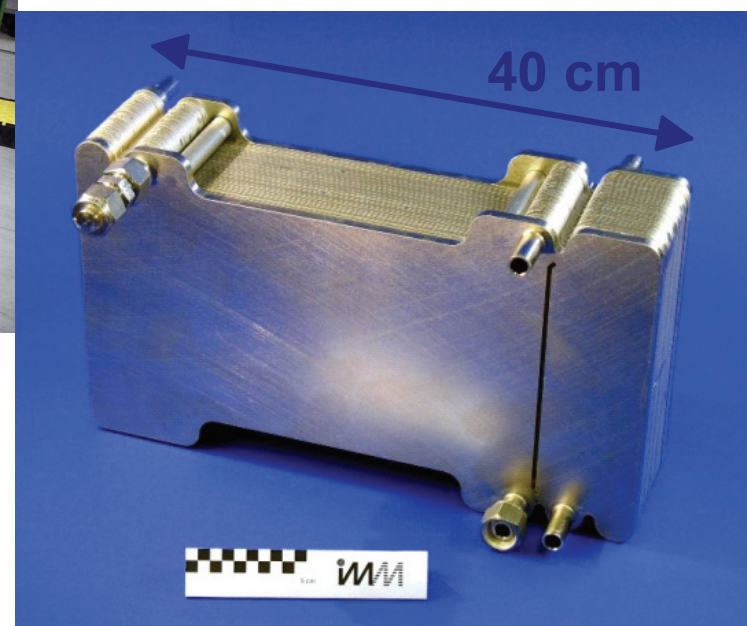
Topical Issue: Micro Process Technology. Guest Editor: Volker Hessel

Brazing



Screen printing for brazing

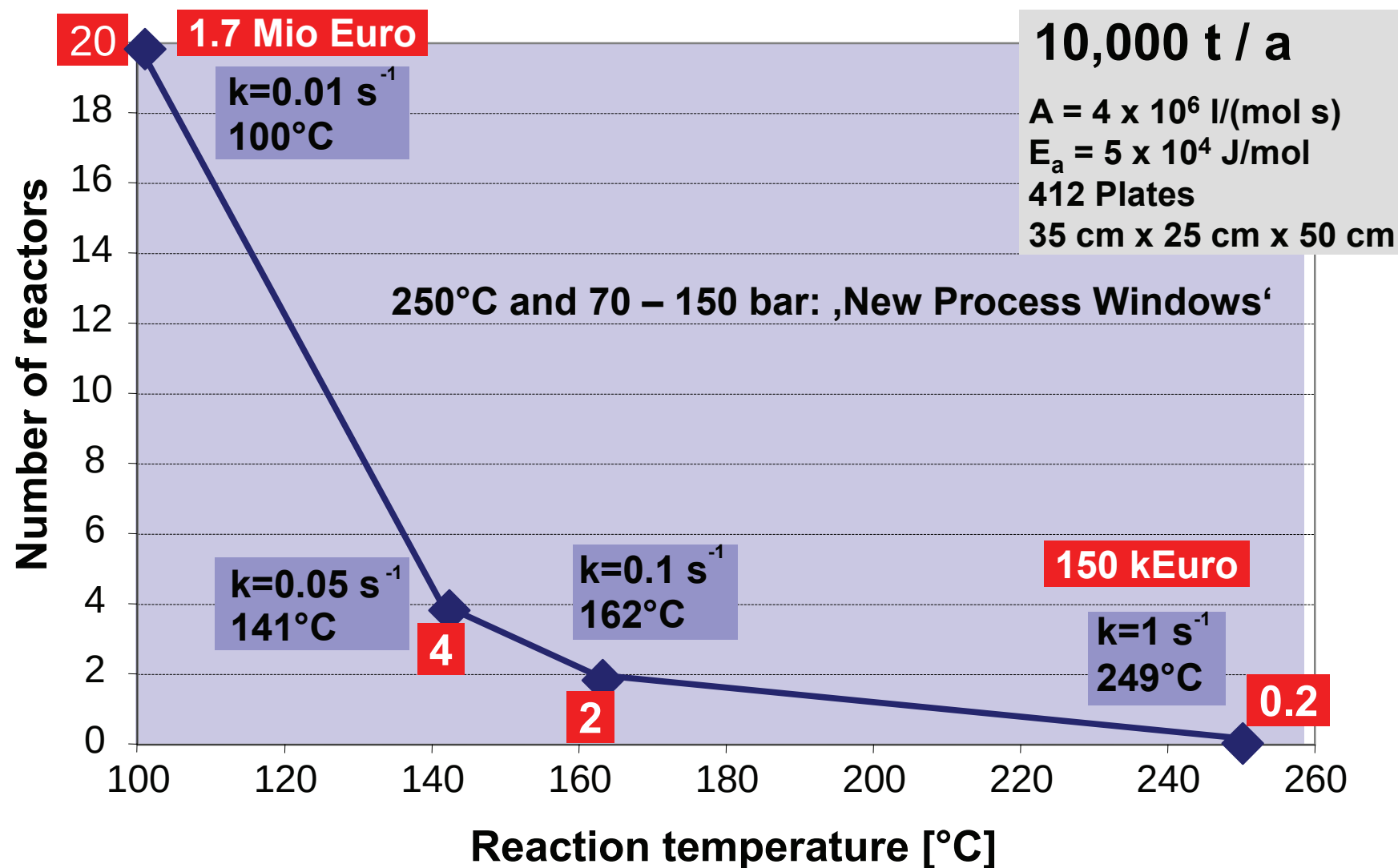
Compact pilot microreactor



Automated catalyst coating line

- **NOVEL PROCESS WINDOWS**
 - From concept to research cluster

HYPOTHETICAL LIQUID-PHASE REACTION AT THE EDGE TO LARGE-SCALE PRODUCTION



NEED TO CHANGE CHEMICAL PROTOCOLS (‘TURN INSIDE OUT’)

{50-130°C; 1 bar; 2-8 h}

Conventional lab batch



{>200°C; >50 bar; <1 s}

High-p,T microreactor protocol

‘alike gas-phase chemistry’

$$k = A \exp(-E_a/RT)$$

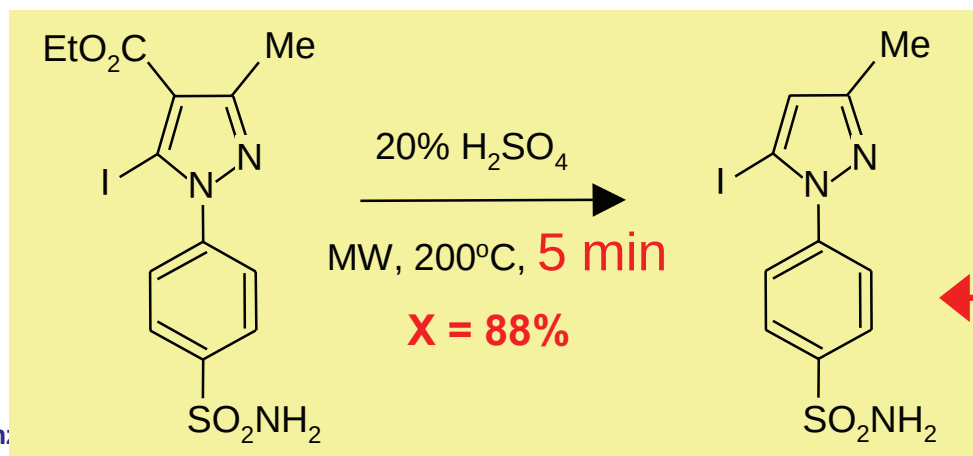
$$A = 4 \times 10^{10} \text{ mol}^{-1} \text{ s}^{-1}$$

$$E_a = 100 \text{ kJ mol}^{-1}$$

Temperature [°C]	27	77	127	177	227
Rate constant k [s ⁻¹]	1.6×10^{-7}	4.8×10^{-5}	3.5×10^{-3}	9.9×10^{-2}	1.43
Time (90% conversion)	68 days	13.4 h	11.4 min	23.4 s	1.61 s

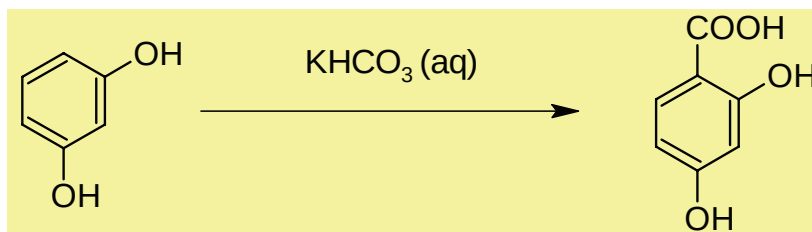
from: Kappe, C.O., Stadler, A. *Microwaves in Organic and Medicinal Chemistry*, Vol. **25** in: *Methods and Principles in Medicinal Chemistry* (eds.: Mannhold, R., Kubinyi, H., Folkers, G.) Wiley-VCH (2005) pp. 94-95.

→ following the routes of microwave assisted continuous synthesis (MAOS)



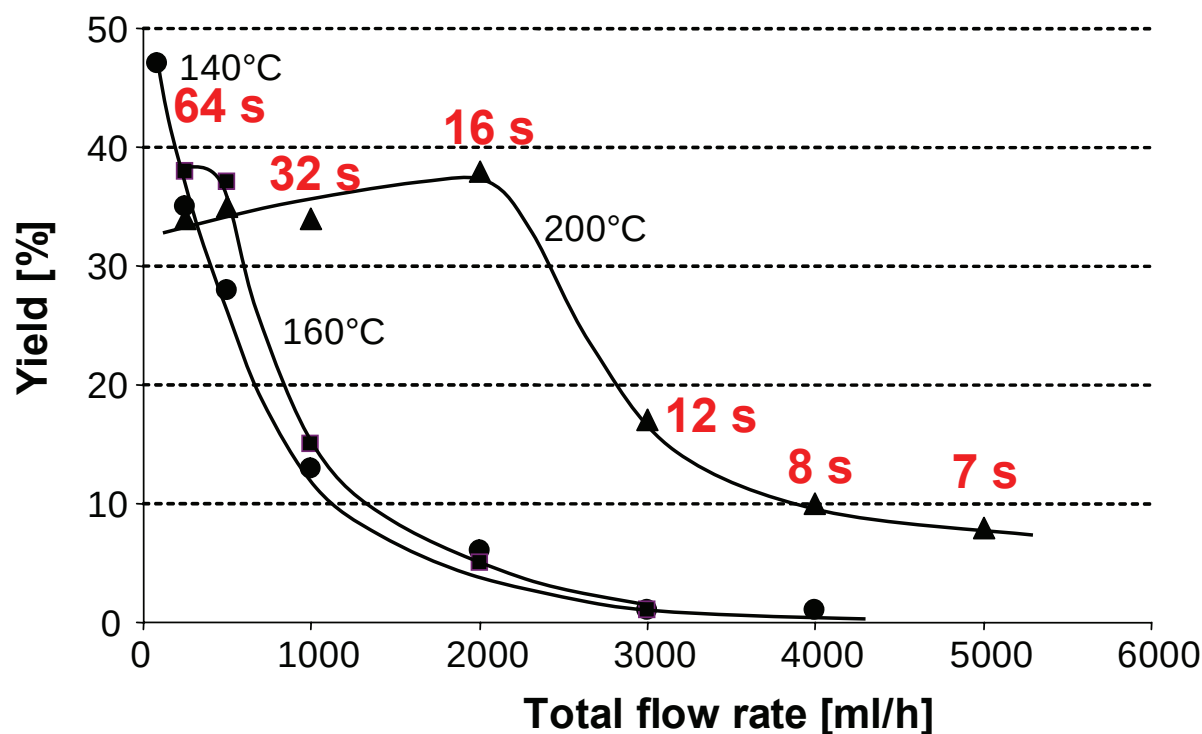
Organ, M.G., Mayer, S., Lepifre, F., M'Zemba, B., Kathri, J. *Mol. Diversity* **7** (2003) 211-227.

Reflux, 100°C, 96 h
 $X = 86\%$



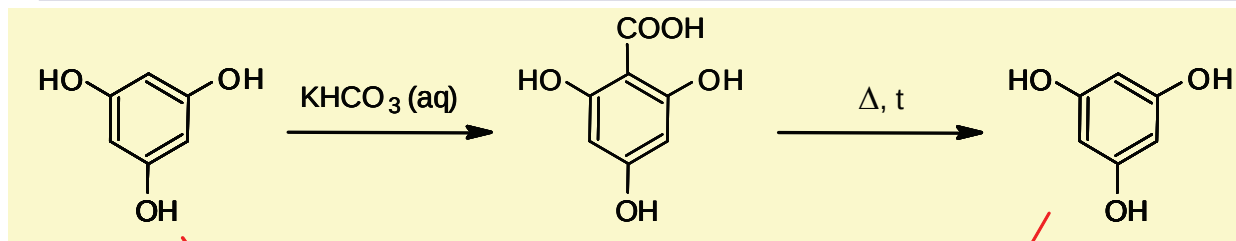
V. Hessel, C. Hofmann, P. Löb, J. Löhndorf, H. Löwe, A. Ziogas
Org. Proc. Res. Dev. **9**, 4 (2005) 479-489

High p,T

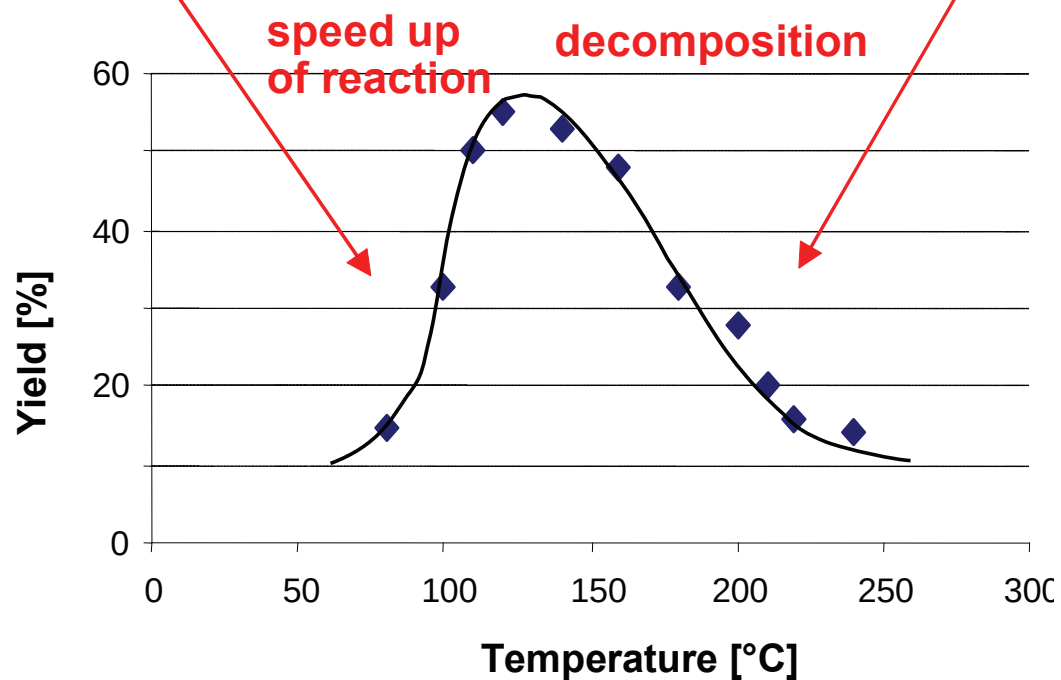


- Pressure: 40-70 bar
- Temperature: 100-220°C
- Reaction time: 4 – 390 s

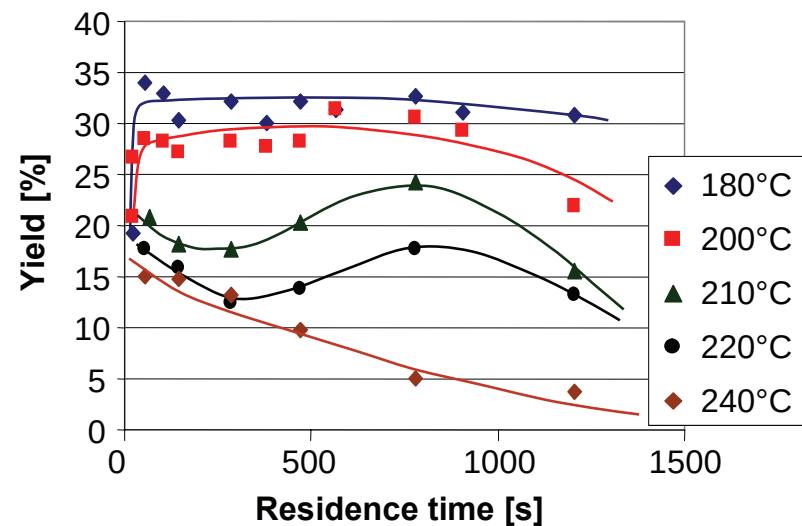
- ➔ Reduction of reaction time by ~2000
- ➔ Increase in space-time yield by factor 440
- ➔ Increase in productivity by factor 4



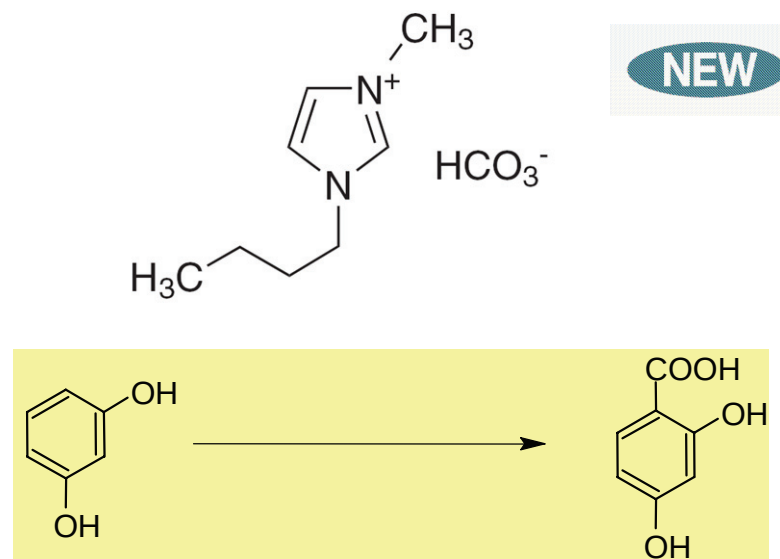
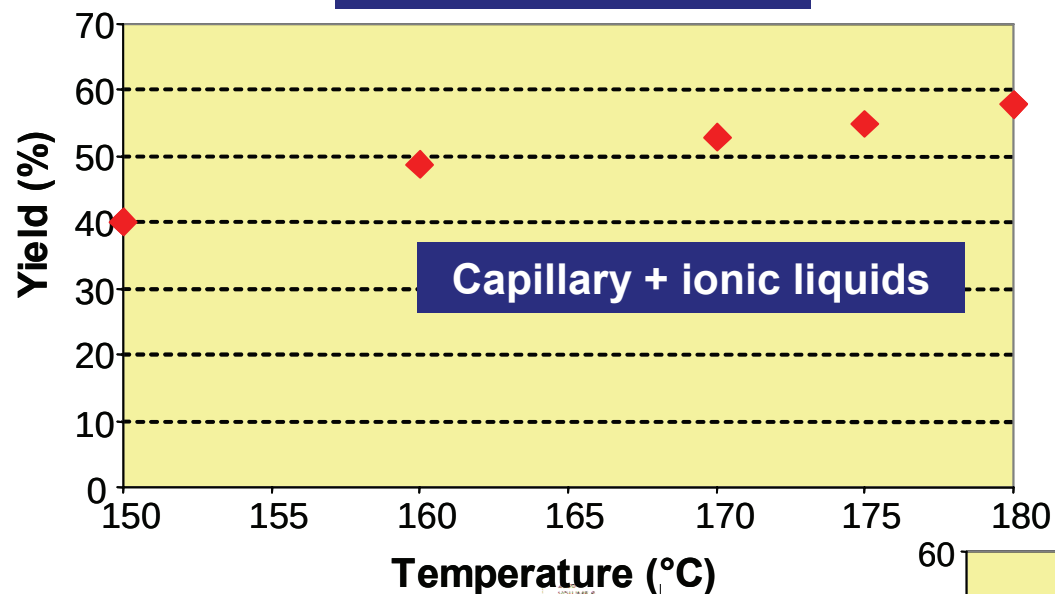
High p,T



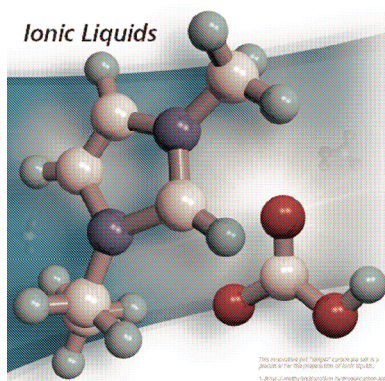
V. Hessel, C. Hofmann, P. Löb, H. Löwe, M. Parals *Chem. Eng. Technol.* 30, 3 (2007) 355-362.



Faysal Benaskar (TUE)
Internship at IMM



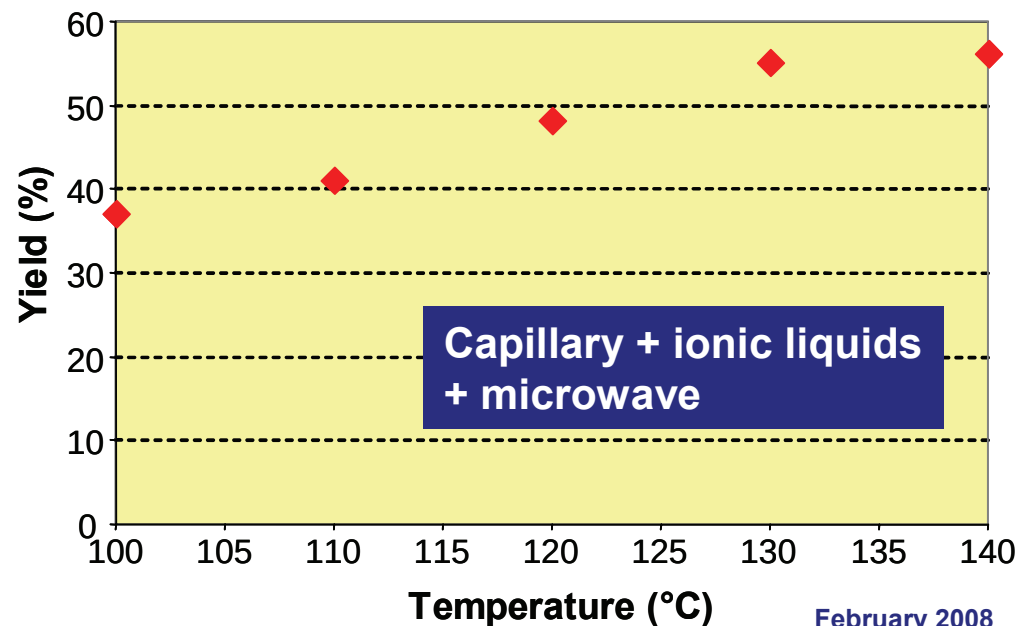
ChemFiles



VOLUME 6
NUMBER 9

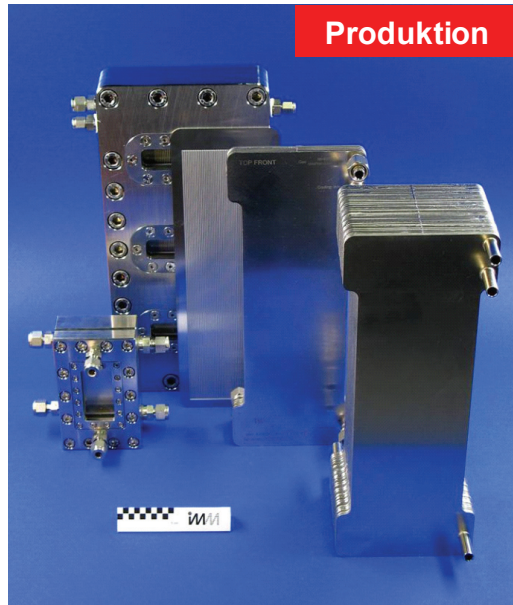
ALDRICH
Fluka

CELLIONIC™
NEW BASONIC™
CBLS®
DIBUTYLPHOSPHATE
AND "NATURAL"
IONIC LIQUIDS
TRIOCTYLMEHYL-
AMMONIUM
THIOSALICYLATE
(TOMATS)
NITRILE (CYANO)
FUNCTIONALIZED
IONIC LIQUIDS



Base case: high-p,T	μ-REAC PLANT – 1 reactor; 4.4 t/a; 10 l/h 45% selectivity; 200°C / 40 bar (,high-p,T')	92.10 € / kg
Selectivity	μ-REAC PLANT – 1 reactor; 4.4 t/a; 10 l/h 70% selectivity; 200°C / 40 bar (,high-p,T')	78.43 € / kg
External numbering-up	μ-REAC PLANT – 10 reactors; 44 t/a; 100 l/h 45% selectivity; 200°C / 40 bar (,high-p,T')	57.47 € / kg
Further PI by high-p,T	μ-REAC PLANT – 1 reactor; 44 t/a; 100 l/h 45% selectivity; 200°C / 40 bar (,high-p,T')	56.95 € / kg
Without high-p,T	μ-REAC PLANT – 1 reactor; 0.01 t/a; 0.025 l/h 45% selectivity; 100°C / 1 bar	17,352.52 € / kg
Bench- marking	BATCH-REACTOR PLANT – 1 l; 0.27 t/a 45% selectivity; 100°C / 1 bar (,Reflux')	985.18 € / kg
Bench- marking	BATCH-REACTOR PLANT – 20 l; 4.3 t/a 45% selectivity; 100°C / 1 bar (,Reflux')	107.05 € / kg

U. Krtschil, V. Hessel, D. Kralisch, G. Kreisel, P. Löb, H. Löwe *Org. Proc. Res. Dev.* (2006) in preparation.
 U. Krtschil, V. Hessel, D. Kralisch, G. Kreisel, M. Küpper, R. Schenk *Chimia* **60**, 9 (2006) 611-617.



291 μ m wide; 106 μ m deep

~300 Euro

Optimum flow rate (ml/min) per channel	Length to reach 95% (m)	Number of channels needed	Number of plate needed	Material cost [%]	Cutting cost [%]	Etching cost [%]	Total cost [%]
0.0222	0.212	751	1	53	20	30	103
0.0109	0.103	1528	2	52	25	30	107
0.0032	0.030	5155	5	51	40	30	121
0.0001	0.001	130363	104	50	535	30	615

1910 μ m wide; 586 μ m deep

Optimum flow rate (ml/min) per channel	Length to reach 95% (m)	Number of channels needed	Number of plate needed	Material cost [%]	Cutting cost [%]	Etching cost [%]	Total cost [%]
30.82	57.02	1	1				
9.73	17.63	2	1				
2.27	3.75	7	1				
0.58	0.87	29	1				
0.18	0.30	92	1	55	20	30	105
0.07	0.14	229	2	70	25	30	125

first comprehensive report: V. Hessel, P. Löb, H. Löwe, *Curr. Org. Chem.* **9**, 8 (2005) 765-787

- **Direct routes from hazardous elements**
- **Routes at increased concentration or even solvent-free**
- **Routes at elevated temperature and/or pressure**
- **Routes mixing the reactants 'all at once'**
- **Routes using unstable intermediates**
- **Routes in the explosive or thermal runaway regime**
- **Process simplification – e.g. routes omitting the need of catalysts or (complex) separation**

**Excellence in Process
Development Research Award**
AIChE (New Orleans / USA, 2008)

compare: Jähnisch, K.; Hessel, V. et al.; *Angew. Chem. Intern. Ed.* **43**, 4 (2004) 406-446

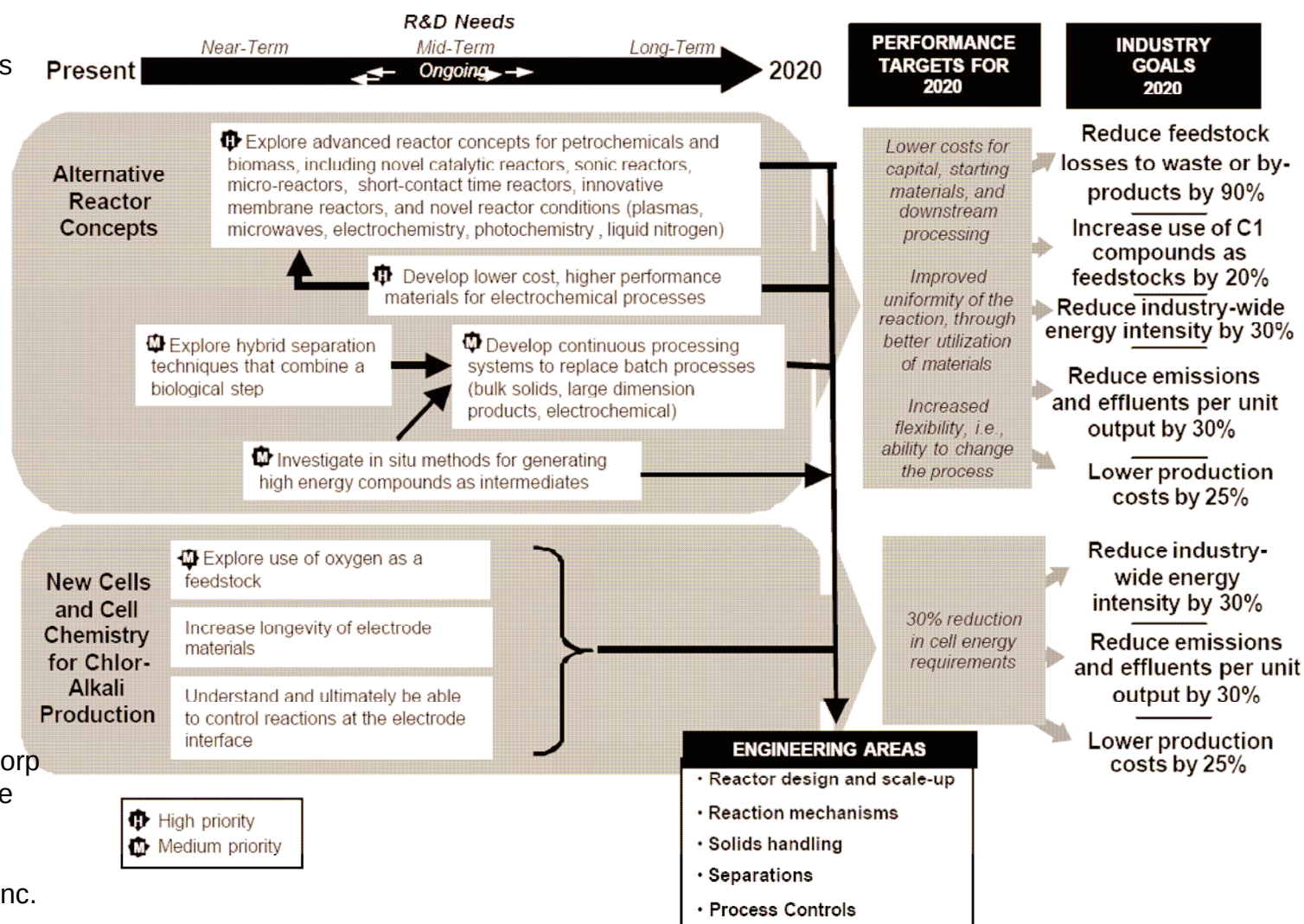
E. R. Murphy, J. R. Martinelli, N. Zaborenko, S. L. Buchwald, K. F. Jensen, *Angew. Chem.* **119** (2007) 1764
F. B. Lopez, PhD thesis, High Pressure: a Challenge for Lab-on-a-Chip Technology, pp.137ff, University Twente (Reinhoudt group)

A.E. Staley Manufacturing
ARCO Chemical
AGA Gas
Air Products & Chemicals
Akzo-Nobel
Autoclave Engineers
BCI
BF Goodrich
Biofine
BOC Gases
Celanese
Cryo Dynamics
Degussa
DOW Chemical
DOW Corning
DuPont
Eastman Chemical
Electrolux
Eltron Research Inc.
Fedegari Autoclavi
Ford Motor Company
Genentech
GE Plastics
Global Technologies
Great Lakes Chemical Corp
Green Chemistry Institute
Isopro International
Itochu Aviation
Kellogg, Brown & Root, Inc.

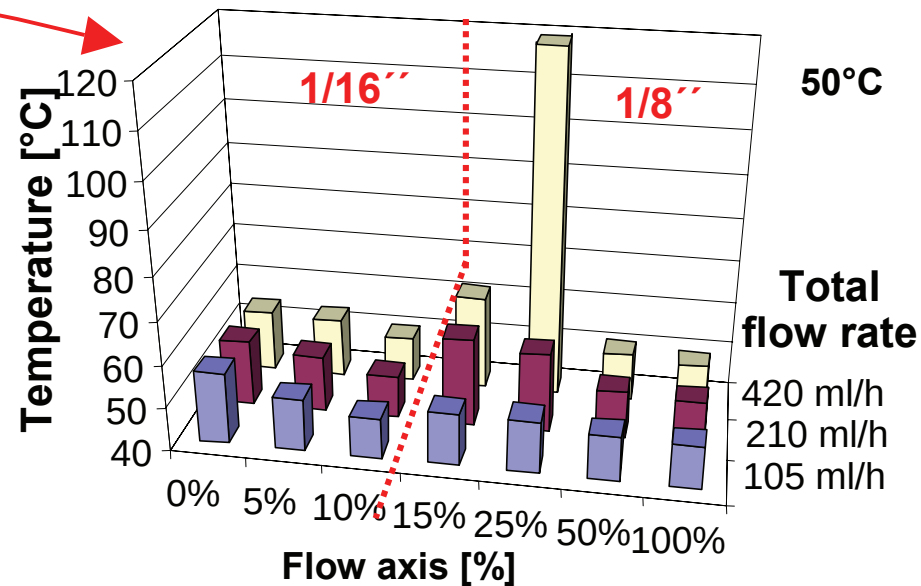
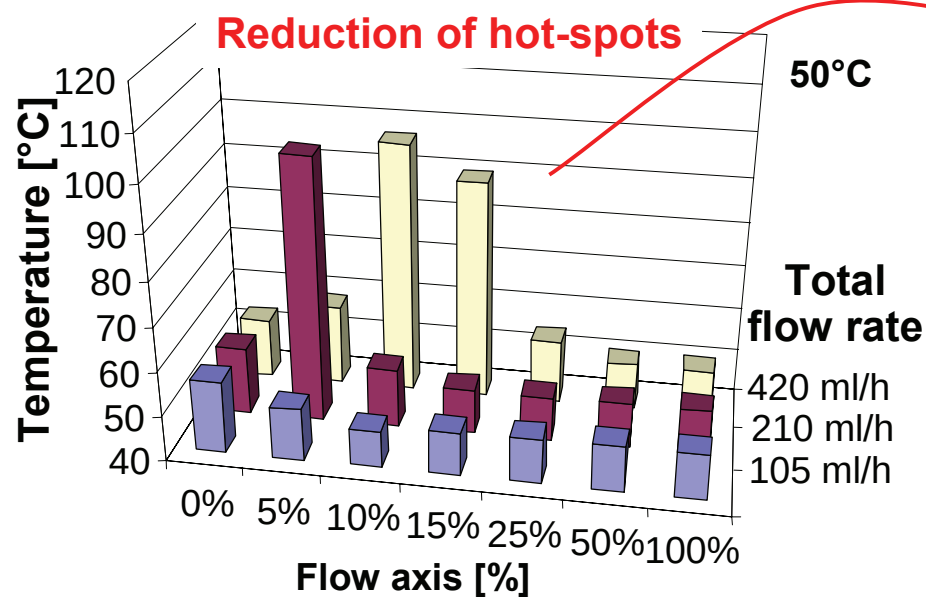
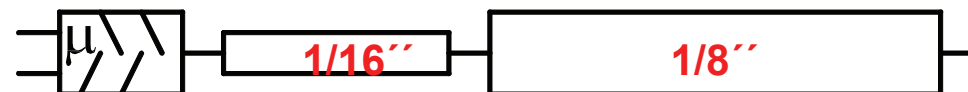
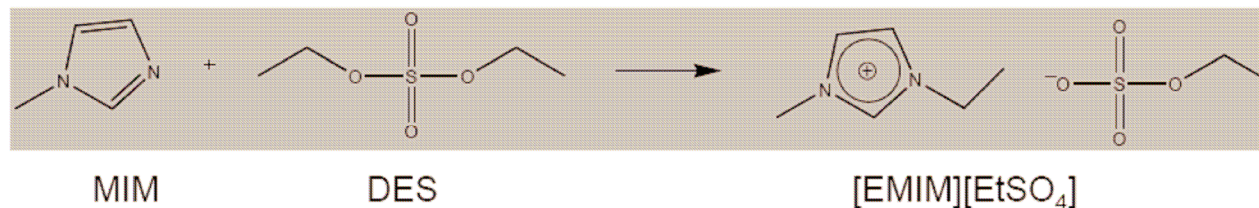
...

hessel@imm-mainz.de

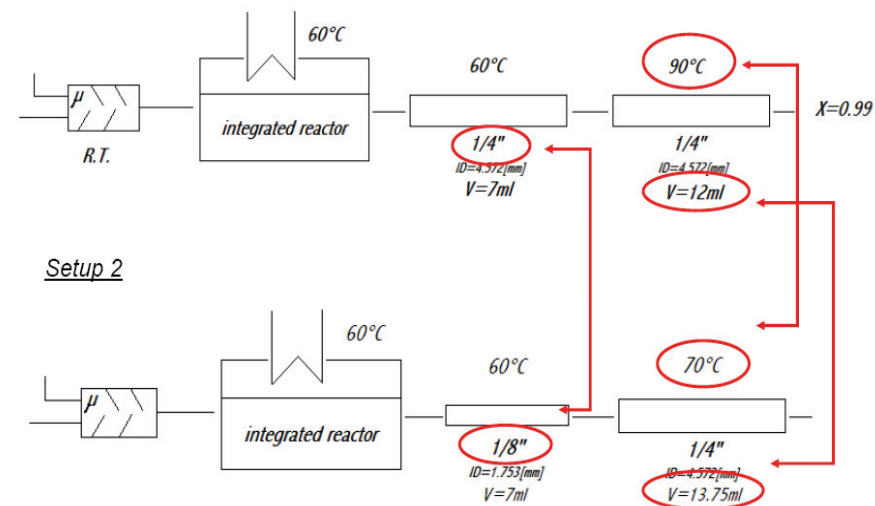
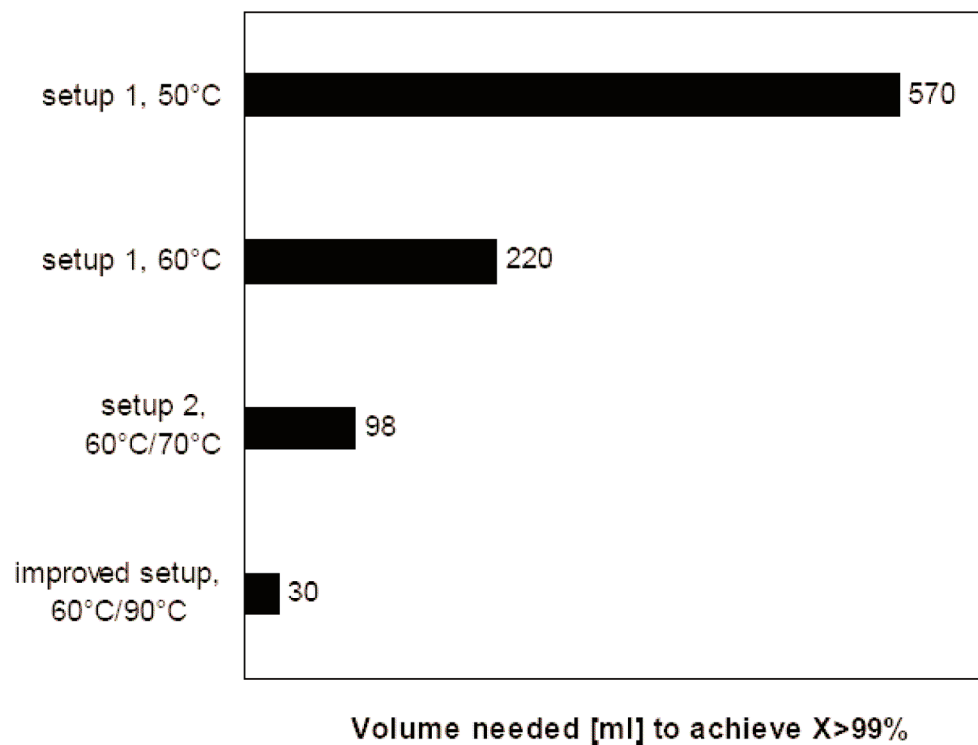
Exhibit 4-5. Process Conditions and Equipment



TWO-SCALED TUBE SECTIONS: IMPROVED TEMPERATURE CONTROL



Prof. Albert Renken, Robert Mischuk (Master)



CONTACTING OF THE DIFFERENT EDUCTS WITH BROMINE (1:1) IN A GLASS MICRO MIXER

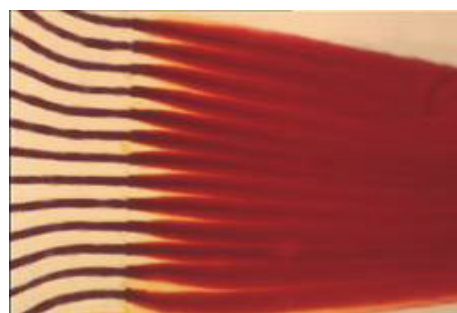
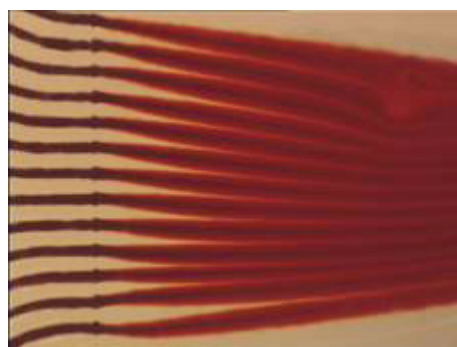
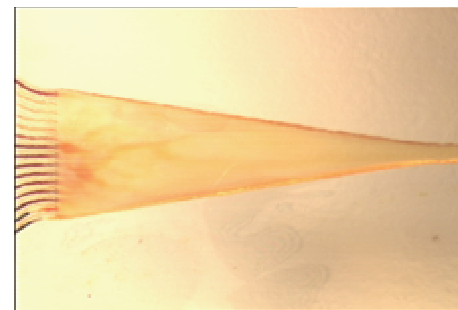
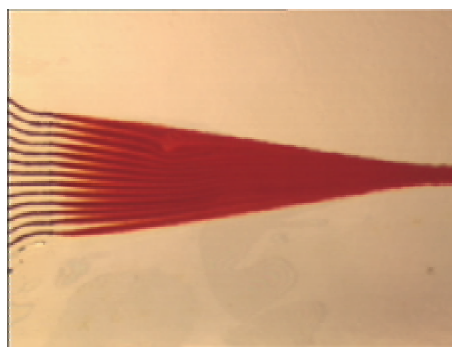
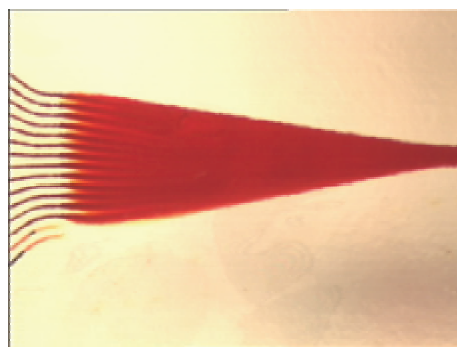
meta-nitrotoluene

toluene

thiophene

reaction speed

formation of gaseous hydrogen bromide



18.6 / 42.8

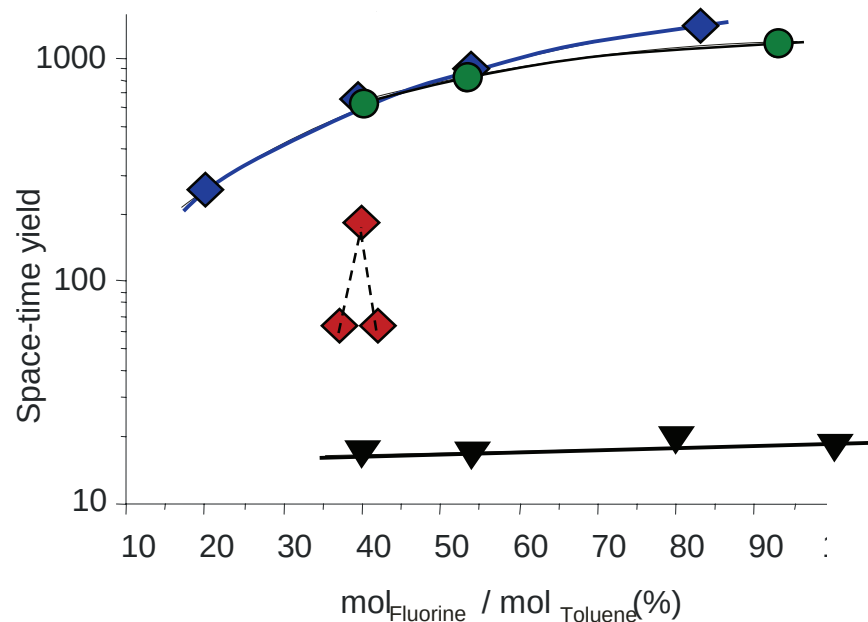
Br₂ [ml/h] / Educt [ml/min]:

24.5 / 49.8

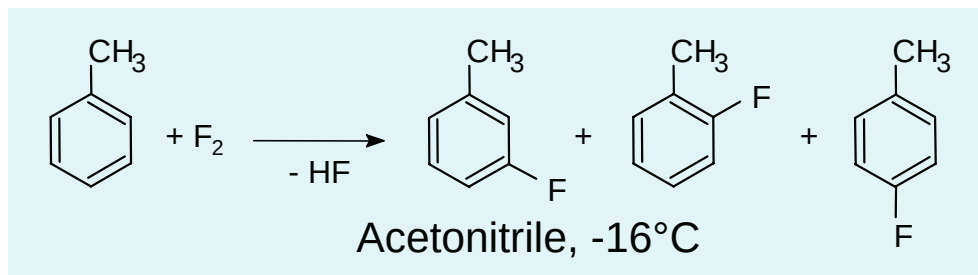
18.4 / 28.4

SPACE-TIME YIELD – THE REDUCTION OF RESIDENCE TIME

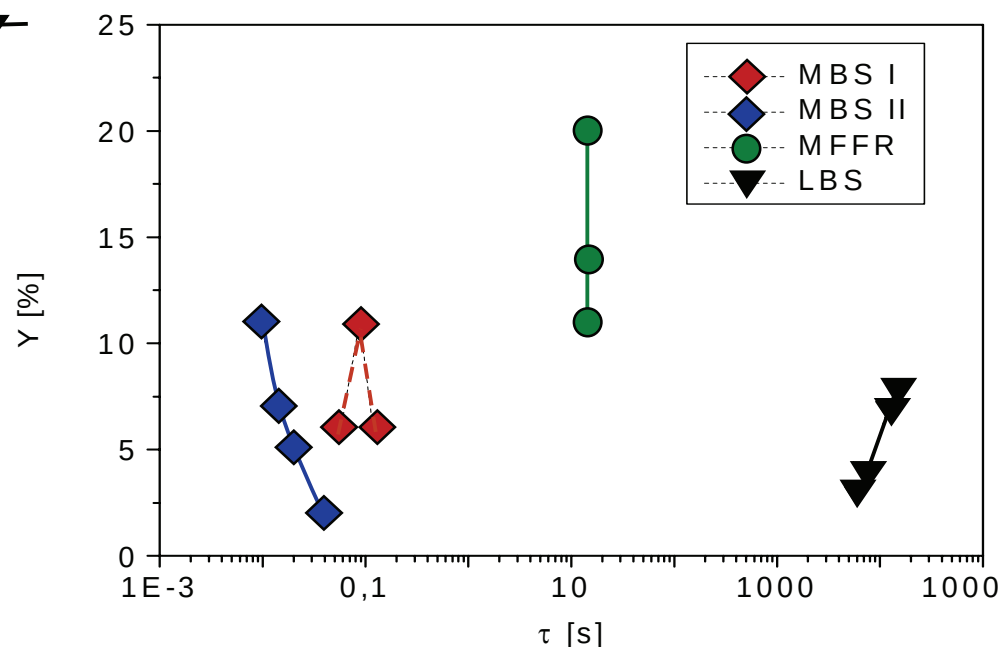
Reaction volume = Active channel volume



- Significant enhancement of space-time yield for the micro-structured reactors compared to the laboratory bubble column



Jähnisch, K., et al.; *J. Fluorine Chem.* **105**, 1 (2000) 117

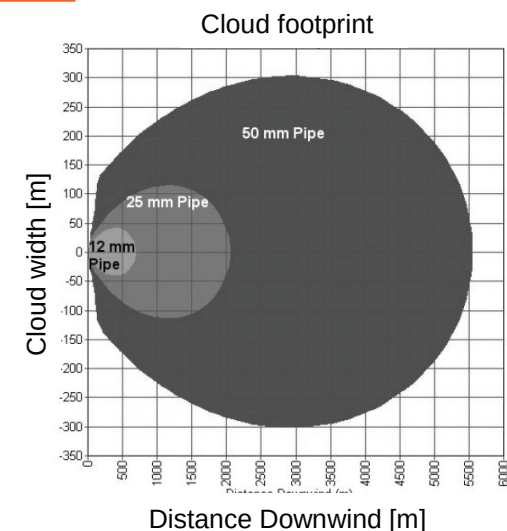


PLACE	DATE	CHEMICAL	ESTIMATED AMOUNT	CASUALTIES
Oppau/Ludwigshafen	September 21, 1921	ammonium sulfate, ammonium nitrate	4,500 t exploded	ca. 550 + 50 dead, 1,500 injured
Flixborough	June 1, 1974	cyclohexane	400 ton inventory, 40 ton escaped	28 dead, 36 + 53 injured
Beek	November 7, 1975	(mainly) propylene	> 10,000 m ³ inventory, 5.5 ton escaped	14 dead, 104 + 3 injured
Seveso	July 10, 1976	2,4,5 trichlorophenol, dioxin	7 ton inventory, 3 ton escaped	no direct casualties, ca. 37,000 people exposed
San Juan, Mexico City	November 19, 1984	LPG	> 10,000 m ³ inventory	5 + ca. 500 dead, 2 + 7000 injured (mainly outside the plant)
Bhopal	December 3, 1984	methyl isocyanate	41 ton released	3,800 dead, 2,720 permanently disabled
Pasadena	October 23, 1989	ethylene, isobutane hexene, hydrogen	33 ton escaped	23 dead, 130-300 injured
Toulouse	September 21, 2001	ammonium nitrate	200-300 ton	31 dead, 2442 injured

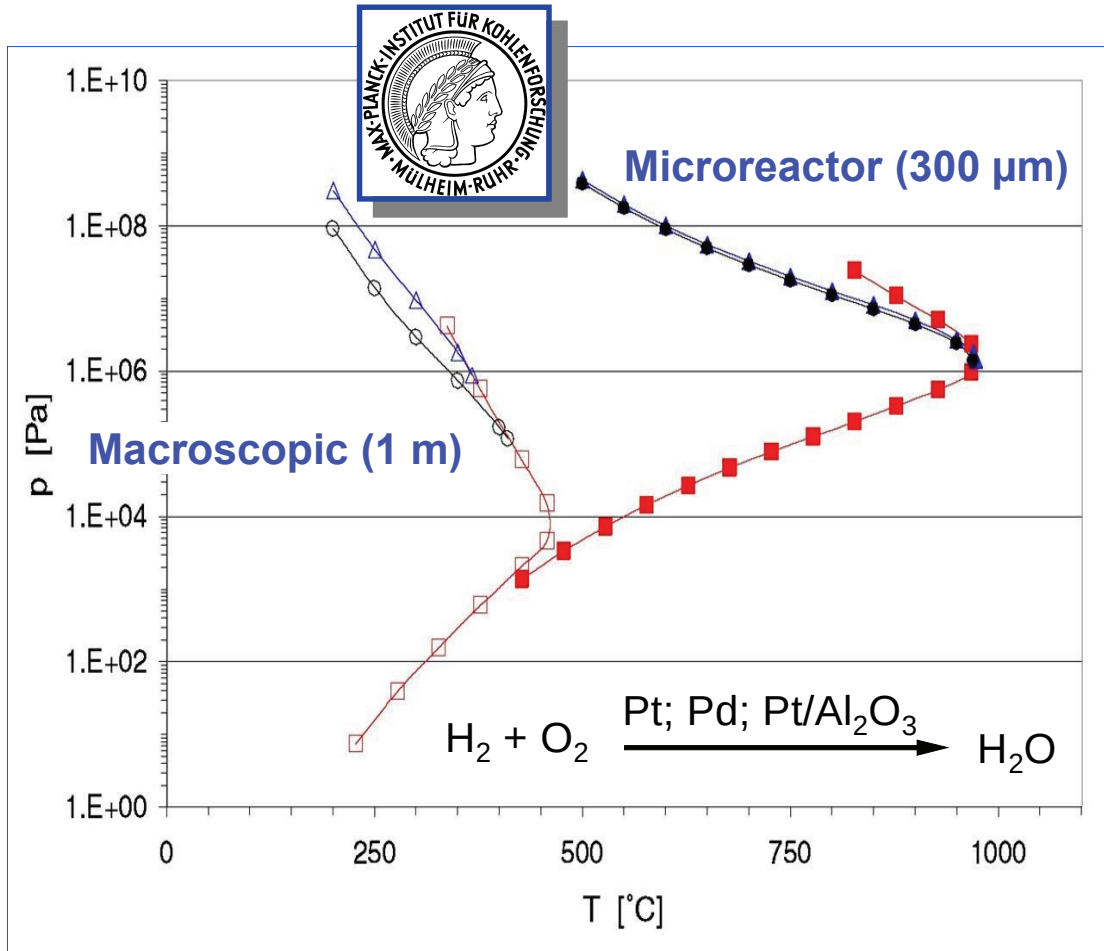
Accidents are due to large inventories !

**UCC, Bhopal 3 Dec 1984:
cloud of 41 tons MIC
>2 500 killed**

**An inventory of 10 kg MIC
would have been sufficient
(D.Hendershot, CEP 2000)**

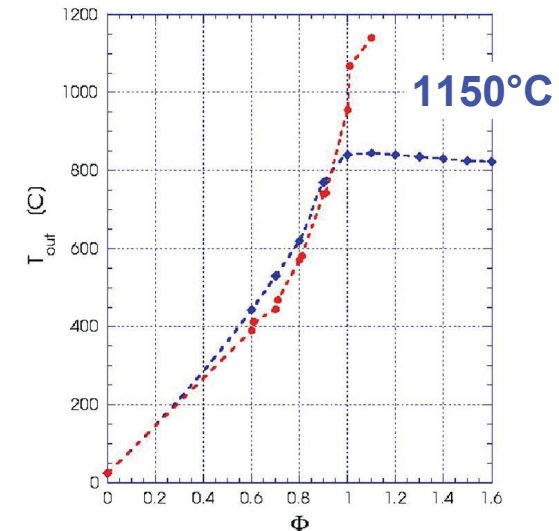


Courtesy A. Stankiewicz (DSM / TU Delft) & J. Jenck (ENKI Tech)



Veser, G.; Chem. Eng. Sci. **56**, (2001) 1265-1273.

Veser, G., Friedrich, G., Freygang, M., Zengerle, R.; *3rd International Conference on Microreaction Technology, IMRET 3*, pp. 674-686, Springer-Verlag, Berlin, (2000).



Explosion limits

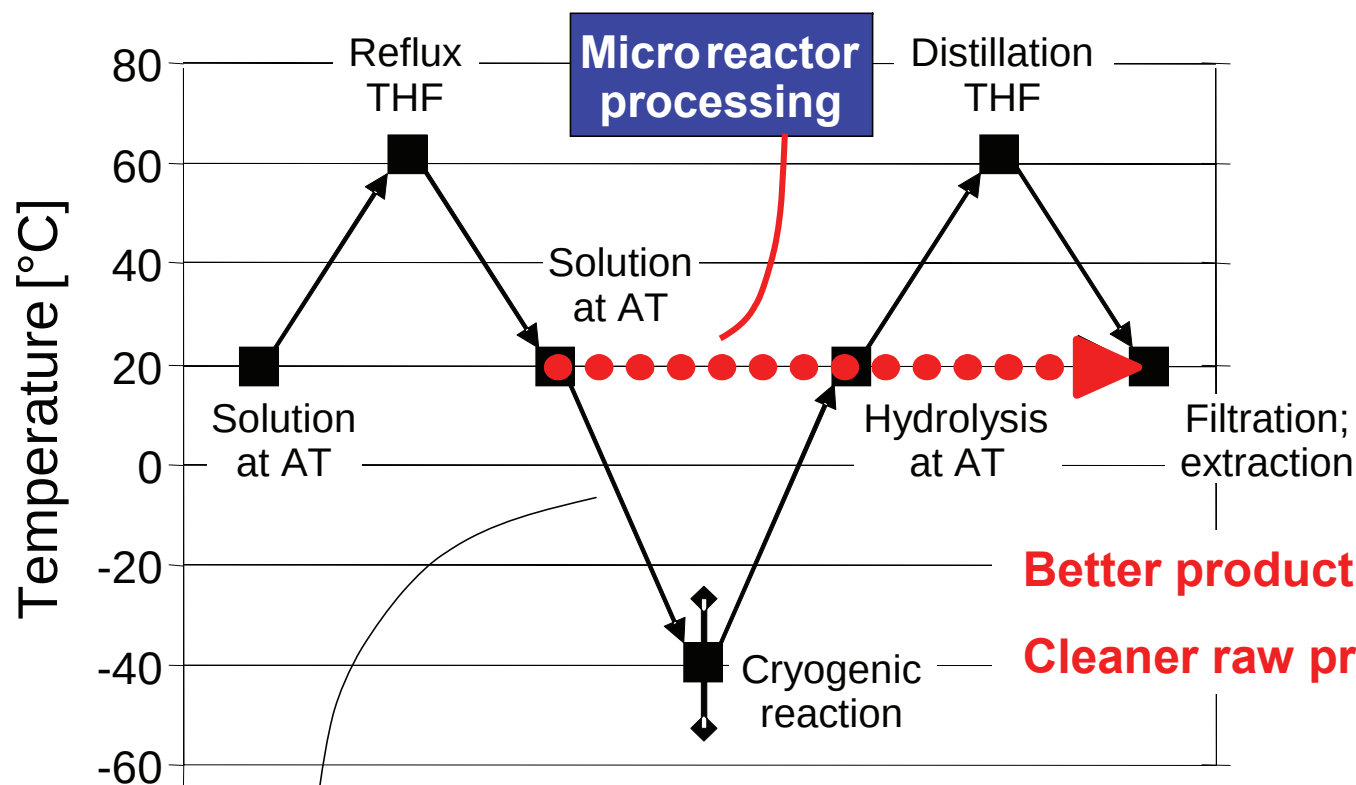
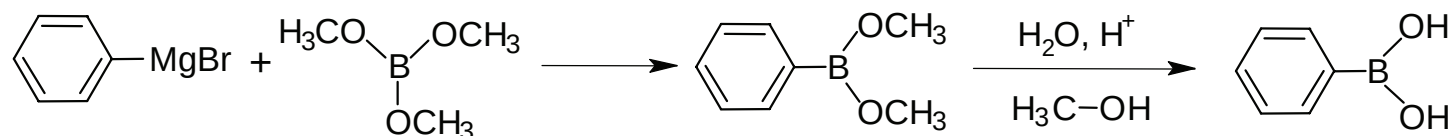
Kinetic limits (squares);
thermal limits (triangles for $T_{\text{wall}} = T_{\text{reaction}}$)
(circles for $T_{\text{wall}} = T_{\text{room}}$)

hessel@imm-mainz.de

Reactor dimensions

- 1) Macroscopic (1 m; open symbols)
- 2) Microreactor (300 μm ; filled symbols)

February 2008



Better product selectivity by 25%

Cleaner raw product by 10%

V. Hessel, C. Hofmann, H. Löwe, A. Meudt,
S. Scherer, F. Schönfeld, B. Werner, *Org.*
Proc. Res. Dev. **8**, 3 (2004) 511-523

Mission

Novel process windows with regard to pressure, temperature, concentration

Sustainable chemical processes by temperature stable, pressurised microreactors

Technical Targets

- **High energy efficiency**
- **Minimisation of waste**
- **Clean and inherently safe product making**
- **Safe and low-emission syntheses**
- **Process intensification (e.g. increase in space-time yield)**

Relevant Topics

- **High-temperature & high-pressure routes, e.g. for functionalisation of alkanes**
- **Syntheses in explosive and thermal runaway regimes**
- **Solvent-less and free syntheses**
- **Multi-step syntheses with immediate conversion of instable intermediates**

New Ways in OLED manufacture

University Jena, POLYMET eV, Jenpolymers Ltd

New process windows for the Kolbe-Schmitt synthesis

IMM, University Jena, Sigma-Aldrich

Preparation of pharmaceutical-grade Chitosan from renewable resources

**Heppe Medical Chitosan GmbH,
University Bremen**

New process windows and manufacturing strategies – batch to conti

University Braunschweig, Pelikan PBS GmbH & CoKG

Ultrasound based catalytic multiphase amination of hydrocarbons at high pressure

Leibnitz-Institut Katalyse, Little Things Factory GmbH, ASD GmbH, JTT

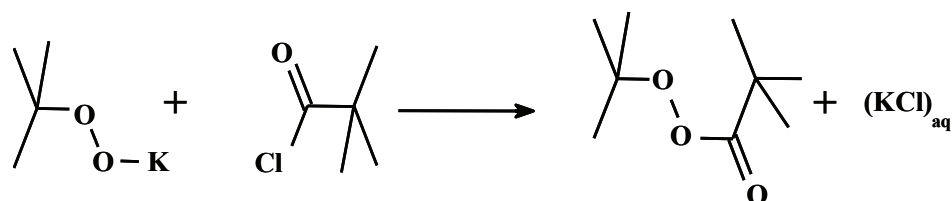
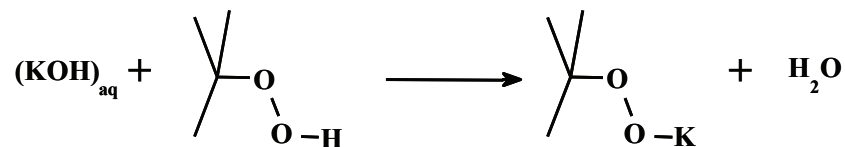
Sustainability evaluation of chemical syntheses during development

University of Applied Sciences Oldenburg

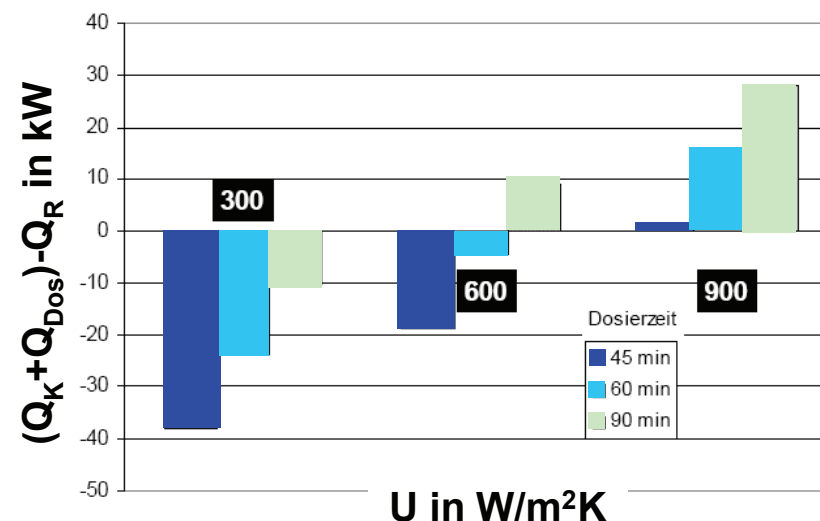


Forschungscluster „Novel Process Windows“





Heat balance of TBPP
synthesis in 1 m³ reactor



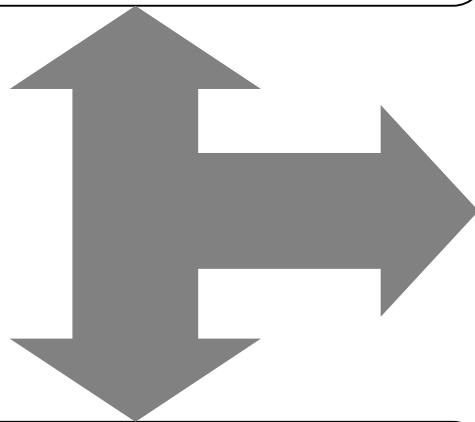
Turnover organic peroxides 1999

120,000 tons, 900 Mio € market value

Explosion of a benzoyl peroxide plant, Catalyst Systems Inc. (Gnadenhutten, Ohio, 2003)

Kralisch, Hübschmann

Entwicklungsarbeit im Rahmen
des Forschungsclusters
„*Novel Process Windows*“



Projektbegleitende
ökologische Bewertung

Novel Process Windows

→ harsche Reaktionsbedingungen (p, T, c hoch)

- p, T → sehr energieintensiv
- c → Sicherheit, mögliche gesundheitliche Gefährdung

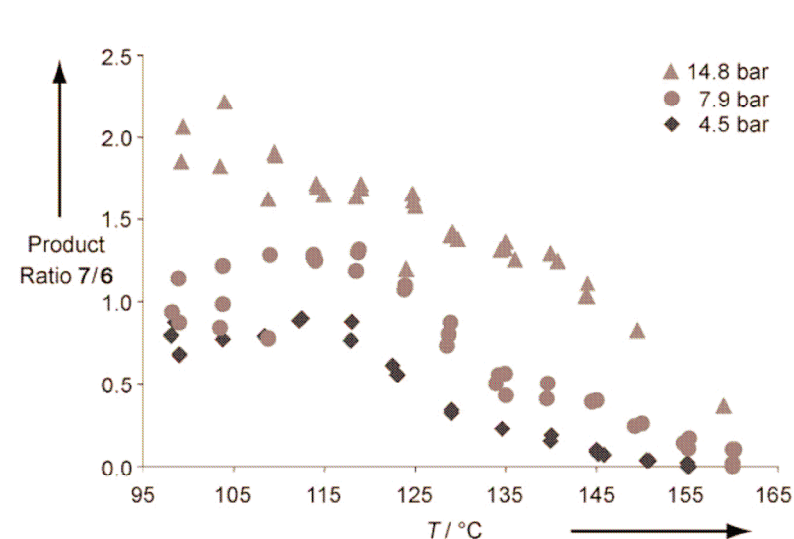
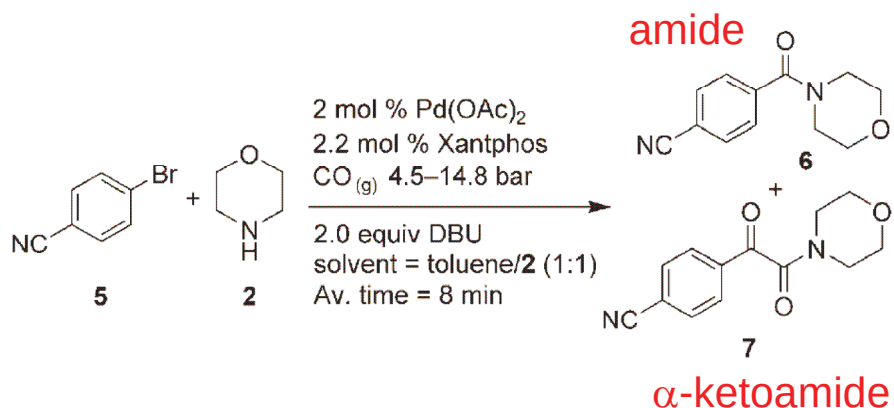
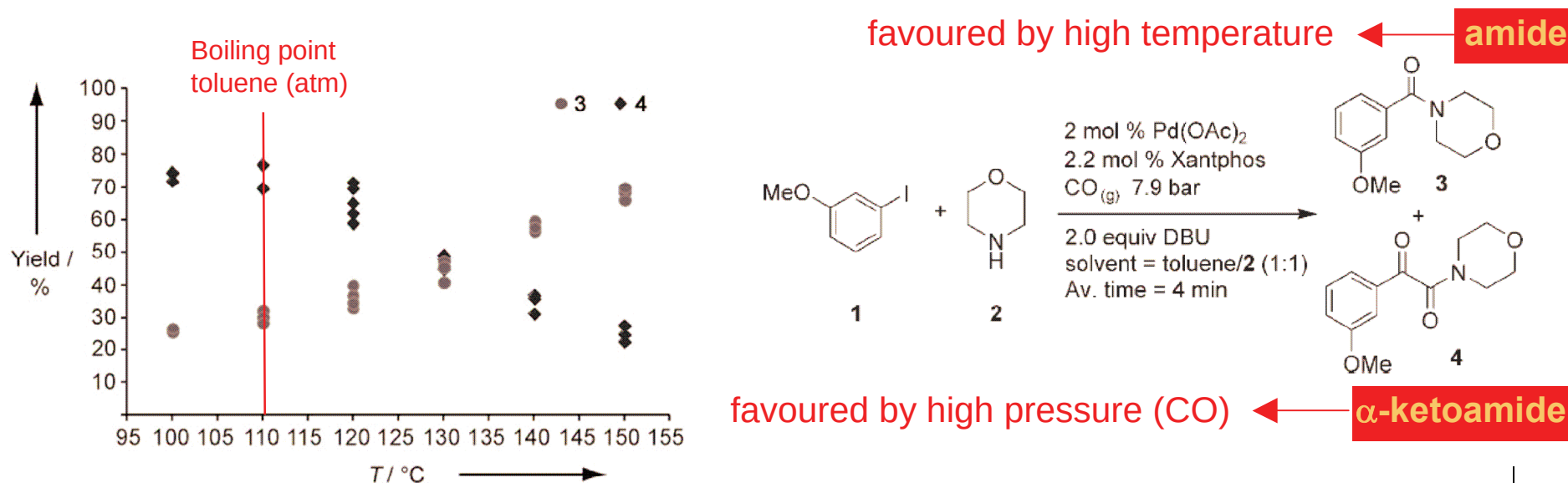
→ Ökologische Bilanz = f(Aufwand/Nutzen)

→ Anforderungen:

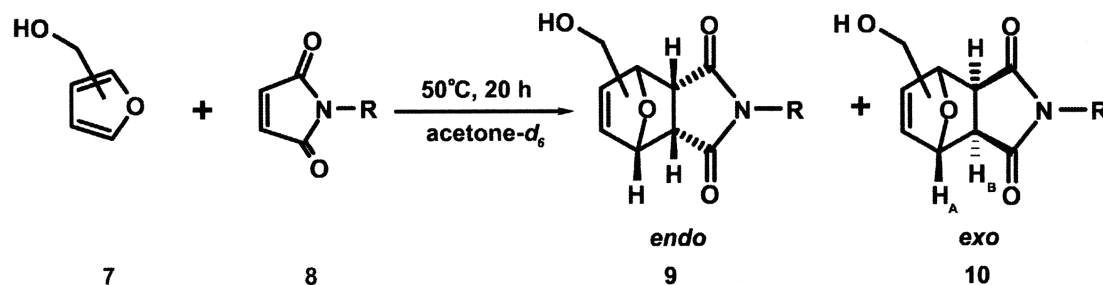
- Signifikant erhöhte Selektivitäten und/ oder Ausbeuten im Vergleich zum Stand der Technik
- Einhaltung von Sicherheitskriterien

- **International Reception**
+ own related activities

Murphy, Martinelli, Zaborenko, Buchwald, Jensen *Angew. Chem. Intl. Ed.* **119**, 10 (2007) 1764-1767



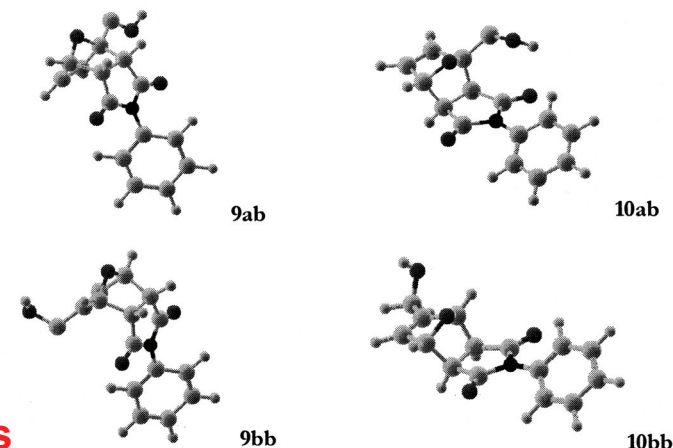
Stereoselectivity of Diels-Alder reactions



a 2-furylmethanol
b 3-furylmethanol

a R = CH₃
b R = Ph
c R = CH₂-Ph

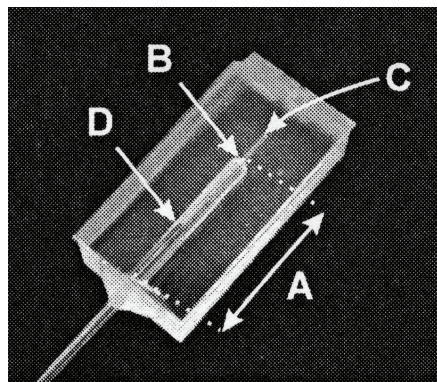
Minimised structures
for reaction volumes ΔV



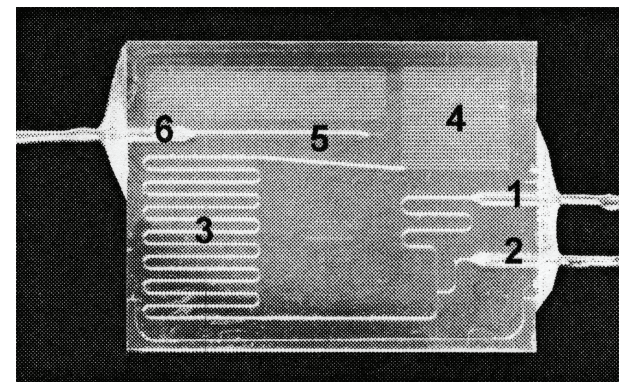
→ Volume difference of endo and exo transition states

7 + 8	Conversion [%]		endo (9) / exo (10) ratio	
	1 bar	600 bar	1 bar	600 bar
aa	35	66	66:34	59:41
ab	35	55	55:45	52:48
ac	40	62	62:38	57:43
ba	57	57	57:43	57:43
bb	55	54	40:60	39:61
bc	59	58	55:45	54:46

Capillary
microreactor

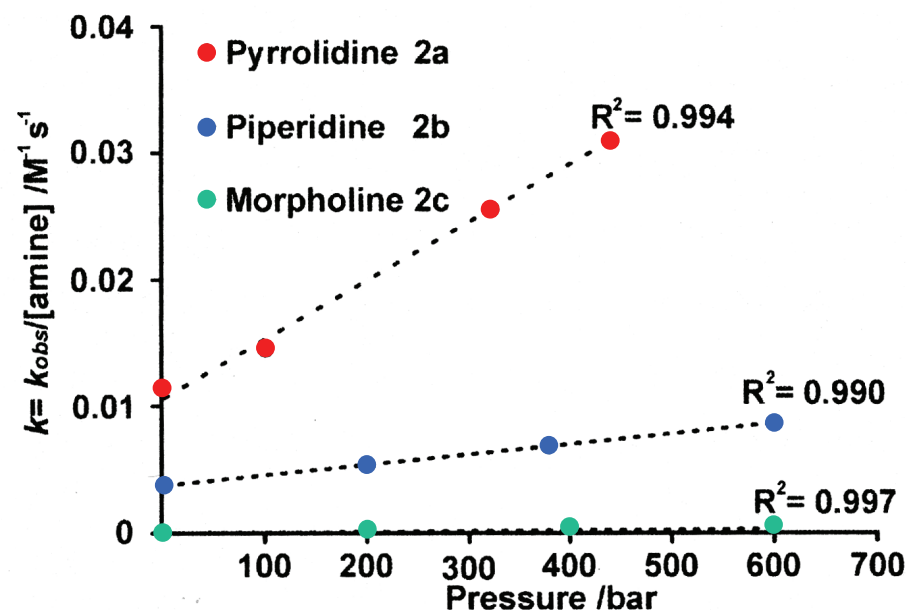
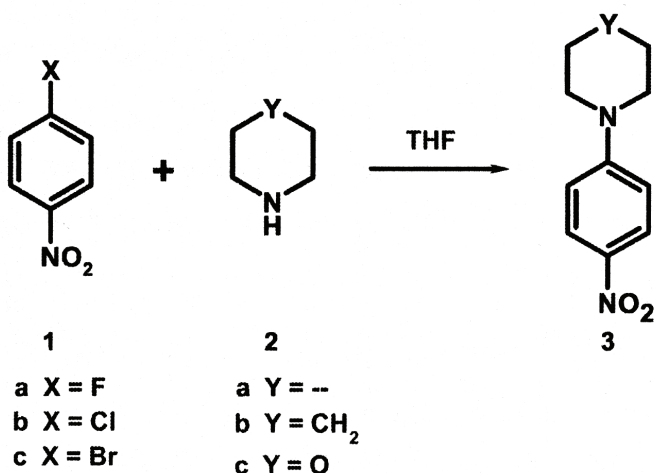


Fiber-based chip interface



Glass high-pressure microreactor

Nucleophilic aromatic substitution



X-CUBE™

High-temperature, high-pressure bench top flow reactor

200°C and 150 bar

Advantages

- Completion of many reactions in minutes
- Optimization "on the fly"
- Low-volume system: conductance of volatile reactions safely and at high pressure

**Suzuki, Sonogashira and Heck couplings,
Buchwald reaction, Halogenation, Alkylation
Dehydration, Esterification, Azide formation**



Heater Cooler

Touch-screen panel Dual pump and injection system

Invited review paper (Hessel, Kralisch) entitled "Sustainability through Greener Processing - New Process Windows by New Processing Technologies"

RSC Publishing [Online Shop](#) [Contact us](#) [Advanced search](#)

Publishing

Home [Publishing](#) [Journals](#) [Energy & Environmental Science](#)

About this Journal

- Journal Home
- About the Journal

Read the Journal

- Advance Articles
- Browse Issues
- Search RSC Journals
- E-Alerts Service
- RSS feed

ReSource

- Submit an Article
- Author Guidelines
- Ethical Guidelines
- Copyright & Permissions
- Referee Report Forms

Explore the Journal

- Hot Articles
- Cover Gallery
- Top 10
- People and Contacts

Customer Services

- Subscriptions
- Advertising
- Librarians

Tools

- ☒ Email this to a friend
- ☒ Email your librarian

Energy & Environmental Science

A new journal linking all aspects of the chemical sciences relating to energy conversion and storage, alternative fuel technologies and environmental science.

Latest News

Read for FREE

Call for Papers

Free online access for new journal
29 January 2008
Registration will provide free institutional access to Energy & Environmental Science in 2008 and 2009

Call for Papers: Energy & Environmental Science
28 January 2008
Energy & Environmental Science welcomes both primary research articles and reviews

New board members in the spotlight
25 January 2008
Energy & Environmental Science is delighted to welcome new members to the Editorial Board

Editorial Board chair announced for Energy & Environmental Science
10 December 2007
Professor Nathan Lewis appointed chair of the Energy & Environmental Science Editorial Board

Signing up to save energy
06 December 2007
Solar powered charger proves popular at the MRS meeting in Boston

New journal announced
26 November 2007

RSC quicklinks

I want information on:

Information for:

I am interested in:

Free to Subscribers

- Chemical Biology
- Chemical Science
- Chemical Technology

Also Of Interest

- Green Chemistry
- Journal of Environmental Monitoring
- Journal of Materials Chemistry
- Physical Chemistry
- Chemical Physics
- Photochemical & Photobiological Sciences

Publish your book with the RSC

Energy and Environmental Science bridges the various disciplines involved with energy.

First issue launched this summer

/department of chemical engineering
and chemistry

TU/e technische universiteit eindhoven

ST Colloquium

Tuesday, 12 February 2008, 4:00 p.m.; Lecture hall STC 0.01

Micro Process Engineering Novel Process Windows in Organic Chemistry and System Design in Fuel Processing

Prof. dr. Volker Hessel

*Institut für Mikrotechnik Mainz (Germany)
Eindhoven University of Technology*

Prof. Dr. Volker Hessel, born 1964, studied chemistry at Mainz University. Topic of his PhD thesis in the field of organic chemistry in 1993 was structure-property relationships of specially designed supra-molecular structures, micelles and lyotropic liquid crystals. Since 1994 Prof. Dr. Hessel is an employee of the Institut für Mikrotechnik Mainz GmbH and in 1996 and 1999 he became group and department leader for microreaction technology, respectively. In 2002 he was appointed as Vice Director R&D and in 2007 as Director R&D. His fields of research comprise micro process engineering for mixing, fine chemistry, fuel processing and heterogeneous catalysis. In July 2005, Prof. Dr. Hessel was appointed as part-time professor for the chair of "Micro Process Engineering" at Eindhoven University of Technology, Chemical Reactor Engineering Group, Department of Chemical Engineering and Chemistry. Prof. Dr. Hessel is author or co-author of 129 peer-reviewed publications (with 22 extended reviews), 11 book chapters, and 3 books. Recently, he received the AIChE award "Excellence in Process Development Research". He is AIChE chair "Microprocess Engineering"; elected board member of the German industrial platform IPμVT; member of five editorial boards (e.g. "Catalysis Today", "Chemical Engineering Journal"); Chair of the program committee for the SynTOP and the IMRET-10 conferences.



Abstract

Micro process engineering has improved organic syntheses many times on the laboratory level. Several case studies from industry at the pilot and production level in fine chemistry were reported and show up robust scale-out concepts with competitive plant & process design. Cost analysis is an integral part here. The key to maximise process intensification is to explore novel process windows – e.g. at high temperatures and pressures ('high-p,T'), by solvent-free processing, or exploration of so far 'hidden' explosive regimes. The combination with other promising technologies like microwaves or ionic liquids is another means to improve performance. Another major application is fuel processing for delivering hydrogen to fuel cells. Fuels under investigation are methanol, (bio-)ethanol, diesel, iso-octane, propane, and methane. Catalyst testing is done using small isothermal microreactors with a few microchannels wash-coated with catalyst. Then, numbering-up is made within one step to shoebox-sized microstructured reactors (1-5 kW_{th}), typically comprising several 10,000 microchannels. This includes integration to microstructured heat exchange units. First, the single reactors are tested in a pilot plant; then the complete process flow in a breadboard configuration. Examples of use will be given, e.g. a complete thermally integrated fuel processor to convert LPG into electricity which is now introduced into the market by industry.

CE&T Special Issue „Microreaction Technology“ March 2007

Chem. Eng. Technol. 2007, 30, No. 3, 289–289

289

Editorial

Novel Tools – Novel Chemistry

In 2008, there will be the first joint ACS and AIChE Conference – chemistry meets chemical engineering and vice versa. This is going to be the largest event in both fields ever and almost 10,000 attendees are anticipated. It took a history of 100 years of AIChE and a similar period of time of ACS to put this idea into practice. Both scientific communities are otherwise more separated and still form own groups in companies with separate contributions at different stages in process development. This is slowly changing recently, as the success of joint efforts have been increasingly recognized.

Micro process technology with its new tools – the microreactors – has become some kind of accepted revolution in chemical engineering. Most chemists – and the author of this Editorial and editor of this special issue of Chemical Engineering & Technology is one of them – have not noticed this so far and, and even if so, they still wonder what to do with such ultra-fast mixing and intensified heat transfer ... except to obtain a higher yield, which seems to be outside common research interests; but the novel tools do not seem to result in new molecules, which is in their focus of interest.

There is something to be discovered by the chemists, though. Essentially hardly any of the traditional synthesis pathways can bring a microreactor to its limit. We are used too much to dilute reactants, to prolong reaction times by extended serial procedures, to avoid usage of aggressive reactants. With the new tool, syntheses can be performed within delays below one second, solvent-free, at high pressure and high temperature, performing all steps simultaneously, having superior process control – this is really making chemistry possible and revolutionizes chemical protocols up to industrial cost analyses. Most of us are not used to actively tune chemistry in the way depicted – a 'novel chemistry' with proven step-change potential is possible now for everyone in the field. The exact consequences are yet not clear yet, but chemistry will not be unaffected.

Thus, chemists should be open about how to use this novel tool for their benefit. A 'Novel Chemistry' steering forum has been initiated just a few weeks ago by the Deutsche Bundesstiftung Umwelt (DBU). Initiatives of the German Engineering Federation VDI, DBU and the German Chemical Industry Association VCI have just supported university clusters to have access to the novel tool and technology so that this facet is added to the education of the young generation. DECHEMA, Society for Chemical Engineering and Biotechnology, organized and hosted the IMRET9 Conference in Potsdam last year and included a workshop on the economic aspects of micro process technology. The VDI-Wissensforum plans to introduce a new conference "Smart Synthesis and Technologies for Organic Processes" early in summer 2008, also combining the two worlds.

The contributions in this special issue origin from the Topical 1 "Applications of Microreactor Engineering" of the AIChE Spring Meeting, April 23–25, 2006 in Orlan-



Volker Hessel
Eindhoven University of Technology –
Chair Micro Process Engineering
AIChE Area Chair Micro Process
Engineering (PDD-12e)
Institut für Mikrotechnik Mainz GmbH

Wednesday, April 9, 2008: 2:00 PM-5:00 PM
Room 392 (New Orleans Convention Center)

**Topical 5: IMRET-10: 10th International Conference on Microreaction
Technology (T5)**

#198 - Novel Process Windows – Unusual Ways of Processing (T5015)

This session deals with the benefits of novel process windows like high temperature and high pressure applications or unusual synthesis routes with high heat releases if they are combined with the advantages offered by microreaction technology.

Chair: [Volker Hessel](#)

CoChair: [Willem Verboom](#)

- 2:00 PM [198a](#) Micro-Reaction Technology (MRT) for the Safe Industrial Application of Diazoalkane-Chemistry: New Chances for Process Intensification
Michael Struempel, Bernd Ondruschka, Annegret Stark
- 2:25 PM [198b](#) Synthesis of 5-Hydroxymethylfurfural from Renewable Resources Using Microreaction Technology
Stefan Loebbecke, Tobias Tuercke, Srinivas Yada, Slobodan Panic, Detlef Schmiedl, Birgit Kamm, Thomas Frank
- 2:50 PM [198c](#) Continuous Microreactor Processing, Reactive Ionic Liquids and Microwave Heating – Approaches towards Intensified Kolbe-Schmitt Synthesis
Ulrich Krtischil, Faysal Benaskar, Volker Hessel, Patrick Löb
- 3:15 PM [198d](#) Effect of Surface-to-Volume Ratio in Microplasma Reactors
Anil Agiral, K. Seshan, Leon Lefferts, J.G.E. (Han) Gardeniers
- 3:40 PM [198e](#) Advantages of on-Site Microreactors from Safety Viewpoint
Eero Kolehmainen, Fatemeh Ebrahimi, Terhi Virkki-Hatakka, Benguang Rong, Ilkka Turunen
- 4:05 PM [198f](#) Reaction Engineering Investigations of the Uncatalysed Cyclohexane Oxidation in A Microstructured Reactor
Thomas Lange, Johannes Fischer, **Elias Klemm**
- 4:30 PM [198g](#) Novel Routes to Fluidized Bed Process Intensification Using a Centrifugal Field
Juray De Wilde

See more of [Topical 5: IMRET-10: 10th International Conference on Microreaction Technology](#)

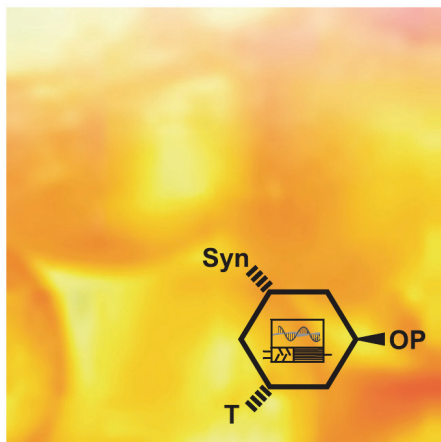
VDI Wissensforum

Conference Announcement and
Call for Papers

1st SynTOP

June 11-13, 2008
Potsdam, Germany

**Smart Synthesis and Technologies
for Organic Processes**
Bridging Chemistry and Engineering



www.SYNTOP2008.com

358852

Main Topics of the Conference

- 1 **Micro and Milli Continuous Process Technologies**
Microfabricated reactors and other devices (plate-type)
Capillary and tube reactors
Foams, monoliths and other structured media
- 2 **Alternative Energy Sources**
Microwaves
Ultrasound
Photochemistry
Electrochemistry
- 3 **Alternative Reaction Media**
Ionic liquids
Supercritical liquids
- 4 **Intensified Plants and Processes**
Intensified equipment
Modular concepts
Case studies
Process development
Piloting
Scale-out
- 5 **Advanced Organic Synthesis & Process Chemistry & Automation**
Multi-step synthesis in micro reactors
Challenging reactions in micro reactors
Case studies

Plenary Lectures



Steven V. Ley, University of Cambridge, United Kingdom
New Tools for Molecule Makers



Michael Matlosz, E.N.S.I.C, Nancy, France
**Microprocess engineering, process intensification
and multiscale design**



Syntop (Ökologie): Lebewesen, die in einem bestimmten Biotop gemeinsam vorkommen

2ND EUCHEMS CHEMISTRY CONGRESS CHEMISTRY: THE GLOBAL SCIENCE SEPTEMBER 16-20 2008 - TORINO, ITALY



RSC | Advancing the Chemical Sciences



**CHAIRMAN
SCIENTIFIC COMMITTEE**
Hartmut MICHEL
(Nobel Laureate,
Frankfurt, Germany)

PLENARY LECTURES BY
Peter AGRE (Nobel Laureate, Baltimore, USA)
Avelino CORMA (Valencia, Spain)
Jean M.J. FRECHET (Berkeley, USA)
Robert H. GRUBBS (Nobel Laureate, Pasadena, USA)
Kyriacos C. NICOLAOU (San Diego and La Jolla, USA)
Martyn POLIAKOFF (Nottingham, UK)
K. Barry SHARPLESS (Nobel Laureate, La Jolla, USA)

Participating Nobel Laureates



Peter AGRE
Plenary Lecture
Nobel Prize in Chemistry in 2003

Johns Hopkins University
School of Medicine
Baltimore, MD, USA

http://nobelprize.org/nobel_prizes/chemistry/laureates/2003/agre-autobio.html



Hartmut MICHEL
Chair of the Scientific Committee
Nobel Prize in Chemistry in 1988

Max-Planck-Institut für Biophysik
Frankfurt am Main, DE

http://nobelprize.org/nobel_prizes/chemistry/laureates/1988/michel-autobio.html



Robert H. GRUBBS
Plenary Lecture
Nobel Prize in Chemistry in 2005

California Institute of Technology (Caltech)
Pasadena, CA, USA

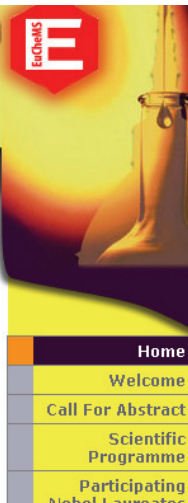
http://nobelprize.org/nobel_prizes/chemistry/laureates/2005/grubbs-autobio.html



K. Barry SHARPLESS
Plenary Lecture
Nobel Prize in Chemistry in 2001

The Scripps Research Institute
La Jolla, CA, USA

http://nobelprize.org/nobel_prizes/chemistry/laureates/2001/sharpless-autobio.html



Home	Advances in Synthesis	Organic Catalysis Convener: J.S. Siegel (CH)	Radical Reactivity in Transition Metal Chemistry Convener: R. Poli (FR)	Reactions under Novel Conditions Convener: V. Hessel (NL/DE)
Welcome	Advances in Understanding	Chemical Measurement Quality: Societal Impact Conveners: B. Magnusson (SE), E. Ferrara (IT)	Cutting Edge Chemistry with Computers Conveners: M. Yáñez (ES), T. Brinck (SE)	Food Safety: Pushing Detection Limits down to Nothing Conveners: R. Battaglia (CH), R. Marchelli (IT)
Call For Abstract	Chemistry and Life Sciences	Biomolecular Interactions and Mechanisms Convener: J. Jiménez Barbero (ES)	Drug Targeting and Delivery Convener: K. Dawson (IE)	Metal Homeostasis Convener: H. Kozłowski (PL)
Scientific Programme	Energy and Industry	Biorefineries and Biotechnologies Convener: M. Penttilä (FI)	Energy Production & Storage Convener: C. Buess-Herman (BE)	New Trends for Agrochemicals Convener: C. Lamberth (CH)
Participating Nobel Laureates	Environment	Greening Chemistry Convener: G. Centi (IT)	Greenhouse Gases Convener: A.A. Jensen (DK)	Water Pollutants Conveners: S. Facchetti (IT), S. Herve (FI)
Other Scientific Topics	Materials and Devices	Branched Polymers - Smart Functional Materials Convener: D.K. Smith (UK)	Nanomaterials Convener: N. Champness (UK)	Porous Materials Convener: U. Müller (DE)
Satellite				
Chemistry the Regis De Accom Prog				

Reactions under Novel Conditions

Convener: V. Hessel (NL/DE)

Keynote:

J.-i. Yoshida (Uni Kyoto)

Topic Lectures:

N. Kockmann (Lonza)

L. Magna (IFP)

H. Lehmann (Novartis)

W. Verboom (Uni Twente)

D. Kralisch (Uni Jena)